

Energy → Pollution → Health

23 March 2007
28th class meeting

READINGS,
Friday 23 March:
None assigned
Saturday 24 March:
Mt Lemmon (see lab website, bring handouts)
Monday 26 Mar:
See Ozone link through EPA



Lab 21/23 Mar:
Energy Lab
(see website or your email)
Saturday 24 March MT LEMMON!
BSE 7am, return 6pm
Lab 28/30 March:
meet in lab

Environmental Biology (ECOL 206)
University of Arizona, spring 2007

Kevin Bonine, Ph.D.
Anna Tyler, Graduate TA

http://eebweb.arizona.edu/courses/Ecol206/206_Page2007.html



Mt. Lemmon Ski Area, March or April 2005

Mt. Lemmon this Saturday...

24 March

Meet 0700h at S side of BSE (corner of 4th and Highland)
"Be on time or get left behind."
Return around 1800h to campus

Please Bring:

Food (snacks and lunch)

We will bring a cooler with ice for you to put stuff in if you wish.

Water (bring a small **bottle**, we will bring big jug for refills)

Hat

Clothes for variable weather conditions. Up top it *might* be in the 30s F and raining or snowing.
Be prepared to enjoy yourself no matter the weather.

Small Notebook, Writing Instruments, Handout from 206 lab website

Sunscreen?
Binoculars?
Camera?
Money?

Please don't bring personal music devices, no text-messaging or phone calls, etc.
We will be interacting as a group all day.



SNR Special Seminar Information

Thanks to financial support from the Graduate Student and Professional Council, Institute for the Study of Planet Earth, and USGS, we are honored to welcome **Dr. Sharon Collinge** from The University of Colorado at Boulder for a visit to SNR on March 27th and 28th. Dr. Collinge will be giving a talk

entitled **Got plague?: Links between landscape change and disease occurrence in western USA grasslands**, as part of the School of Natural Resources Seminar Series (Wednesday, March 28th at noon in the Old Chem Bldg (across the quad) room 209).

Dr. Collinges current research focuses on the role of landscape spatial heterogeneity in shaping ecological processes, including responses of individual organisms, populations, and communities to spatial variation in landscape structure. More information on Dr. Collinges lab can be found at <<<http://www.colorado.edu/eeb/EEBprojects/CollingeLab/index.html>>>.

A selection of Dr. Collinges papers can be found by following this link (<http://www.u.arizona.edu/~obrien/research/collinge/collinge.html>).

Tuesday, March 27th

Undergraduate student luncheon, 12:00-1:00 pm, room 218 BSE

Break-out session (professional development and research), 1:15-2:15, room 218 BSE

Wednesday, March 28th

SNR Noon Seminar, 12:00-1:00, Old Chem Bldg room 209

Potluck social, 6:00-9:00 (more information to come)

March 20, 2007, NYTimes

Material Shows Weakening of Climate Reports

By ANDREW C. REVKIN and MATTHEW L. WALD

WASHINGTON, March 19 — A House committee released documents Monday that showed hundreds of instances in which a White House official who was previously an oil industry lobbyist edited government climate reports to play up uncertainty of a human role in global warming or play down evidence of such a role. In a hearing of the House Committee on Oversight and Government Reform, the official, Philip A. Cooney, who left government in 2005, defended the changes he had made in government reports over several years. Mr. Cooney said the editing was part of the normal White House review process and reflected findings in a climate report written for President Bush by the National Academy of Sciences in 2001.

They were the first public statements on the issue by Mr. Cooney, the former chief of staff of the White House Council on Environmental Quality. Before joining the White House, he was the "climate team leader" for the American Petroleum Institute, the main industry lobby.

He was hired by Exxon Mobil after resigning in 2005 following reports on the editing in The New York Times. The White House said his resignation was not related to the disclosures.

Mr. Cooney said his past work opposing restrictions on heat-trapping gases for the oil industry had had no bearing on his actions once he joined the White House. "When I came to the White House," he testified, "my sole loyalties were to the president and his administration."

Mr. Cooney, who has no scientific background, said he had based his editing and recommendations on what he had seen in good faith as the "most authoritative and current views of the state of scientific knowledge."

Mr. Cooney was defended by James L. Connaughton, chairman of the environmental council and his former boss. The hearing was part of an investigation, begun under the committee's Republican chairman last year, into accusations of political interference in climate science by the Bush administration. It became a heated and largely partisan tug of war over the appropriate role of scientists and political appointees in framing how the government conveys information on global warming.

The hearing ...

March 20, 2007, NYTimes

Material Shows Weakening of Climate Reports

By ANDREW C. REVKIN and MATTHEW L. WALD

The hearing also produced the first sworn statements from George C. Deutsch III, who moved in 2005 from the Bush re-election campaign to public affairs jobs at NASA. There he warned career press officers to exert more control over James E. Hansen, the top climate expert at the space agency.

Testifying at the hearing, Dr. Hansen said editing like that of Mr. Cooney and efforts to limit scientists' access to the news media and the public amounted to censorship and muddled the public debate over a pressing environmental issue. "If public affairs offices are left under the control of political appointees," he said, "it seems to me that inherently they become offices of propaganda."

Republicans criticized Dr. Hansen for what they described as taking political stances, for spending increasing amounts of time on public speaking and for accepting a \$250,000 Heinz Award for environmental achievement from the Heinz Family Philanthropies, run by Teresa Heinz Kerry, the wife of Senator John Kerry, Democrat of Massachusetts.

Representative Darrell Issa, Republican of California, proposed that Dr. Hansen, by complaining about efforts to present two sides on global warming research, had become an advocate for limiting the debate.

Dr. Hansen replied, "What I'm an advocate for is the scientific method."

Mr. Deutsch said his warnings to other NASA press officials about Dr. Hansen's statements and news media access were meant to convey a "level of frustration among my higher-ups at NASA."

Mr. Deutsch resigned last year after it was disclosed that he had never graduated from Texas A&M University, as his résumé on file at NASA said. He has since completed work for the degree, he said Monday. Democrats focused on fresh details that committee staff members had compiled showing how Mr. Cooney made hundreds of changes to government climate research plans and reports to Congress on climate that raised a sense of uncertainty about the science.

The documents "appear to portray a systematic White House effort to minimize the significance of climate change," said a memorandum circulated by the Democrats under the committee chairman, Representative Henry A. Waxman of California.

Speth
 Red Sky at Morning
 Chapter 2, Lost in Eden

Biologists

Harold Mooney and Paul Ehrlich correctly observe that “ignorance of the services that natural ecosystems supply to the human enterprise—of the reasons that the economy is a wholly owned subsidiary of those systems—amounts to a condemnation of schools, colleges, universities and the print and electronic media.”¹¹ This area is one of many where we are paying a high price for our neglect of environmental education at all grade levels.

p.27

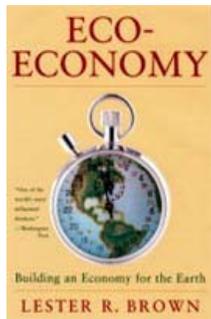
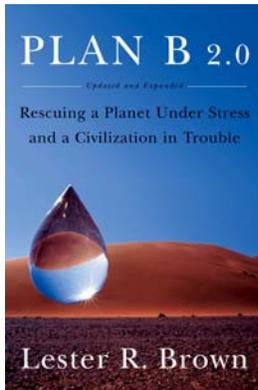
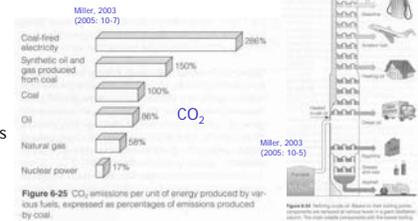
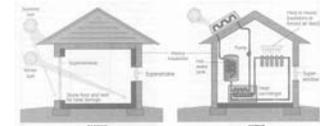
Energy Sources

Solar – passive or active
 Hydrogen - (green algae??)
 need to decompose water

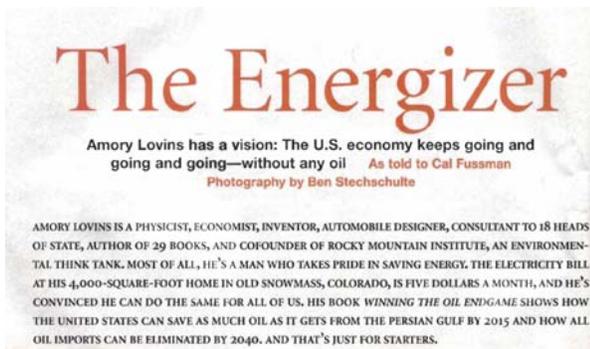
Hydropower
 Wind
 Biomass
 Geothermal

Fossil Fuels
 Oil
 Natural Gas
 Coal

Nuclear



<http://www.earth-policy.org/Books/index.htm>
<http://www.earth-policy.org/Books/index.htm>



WIND

IF I COULD DO JUST ONE THING TO SOLVE OUR ENERGY PROBLEMS, I WOULD allow energy to compete fairly at honest prices regardless of which kind it is, what technology it uses, how big it is, or who owns it. If we did that, we wouldn't have an oil problem, a climate problem, or a nuclear proliferation problem. Those are all artifacts of public policies that have distorted the market into buying things it wouldn't otherwise have bought because they were turkeys.

ELECTRICITY

A QUESTION I ASK A LOT IS, WHAT'S THE RIGHT SIZE FOR THE JOB? I HAVE a book called *Small Is Profitable: The Hidden Economic Benefits of Making Electrical Resources the Right Size*. It points out 207 benefits of distributed resources, such as solar and wind power. When I begin to describe them, you'll find them really obvious:

ENERGY

Until then, the energy problem was generally considered to be: Where do we get more energy? People were preoccupied with where we could get more energy of any kind, from any sources, for any price—as if all our needs were the same. I started instead at the other end of the problem: What do we want the energy for? You don't generally want lumps of coal or barrels of sticky black goo. You want comfort, illumination, mobility, baked bread, and

OIL

LET'S START WITH ONE BASIC PROBLEM. SAUDI ARABIA HAS A QUARTER OF the world's oil reserves. It is the sole swing producer with significant capacity to increase output, and therefore it controls the world price.

DEFENSE

A MAJOR PLAYER IN OUR ENERGY FUTURE WILL BE THE PENTAGON. HERE'S why: Trailing behind every half-mile-a-gallon Abrams tank—a peerless fighting machine if you can get it there—are two un-armored fuel trucks. Guess what the bad guys shoot at?

WEIGHT

A MODERN CAR, AFTER 120 YEARS OF DEVOTED ENGINEERING EFFORT SINCE Gottlieb Daimler built the first gasoline-powered vehicle, uses less than 1 percent of its fuel to move the driver. How does that happen?

Well, only an eighth of the fuel energy reaches the wheels. The rest of it is lost in the engine, drivetrain, and accessories, or wasted while the car is idling. Of the one-eighth that reaches the wheels, over half heats the tires on the road or the air that the car pushes aside. So only 6 percent of the original fuel energy accelerates the car. But remember, about 95 percent of the mass being accelerated is the car—not the driver. Hence, less than 1 percent of the fuel energy moves the driver. This is not very gratifying.

PLASTIC

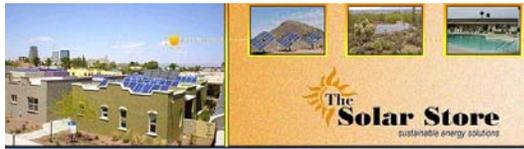
HENRY FORD SAID YOU DON'T NEED WEIGHT FOR strength. If you did need weight for strength, your bicycle helmet would be made of steel, not carbon fiber. And if you want to know how strong a very light material can be, try eating an Atlantic lobster with no tools.

HYDROGEN

MANY AUTOMAKERS ARE STARTING TO UNDERSTAND THAT WHOEVER GOES ultralight first will take the lead in the hydrogen fuel-cell race.



<http://www.ases.org/index.htm>



2833 N Country Club Rd
Tucson, AZ 85716
Phone: 520-322-5180
FAX: 520-322-9531
Toll Free: 877-264-6374

<http://www.solarstore.com/>



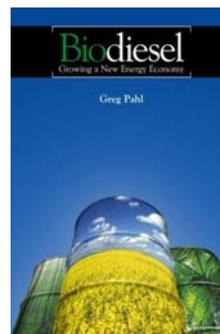
Windfarm off coast of Denmark

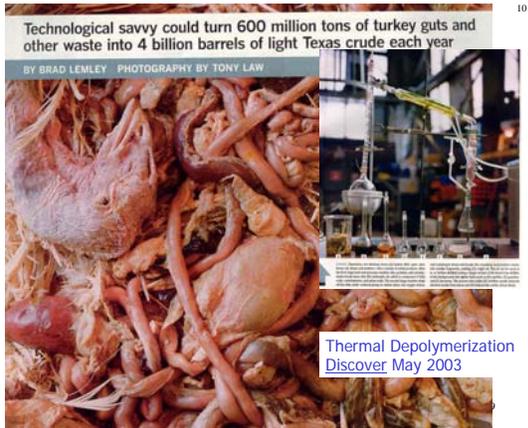


Wave Energy



BioMass and Biofuels





Thermal Depolymerization
Discover May 2003



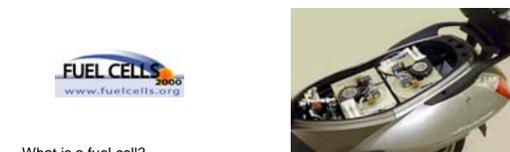
Thermal Depolymerization
Discover May 2003



Ethanol

E85

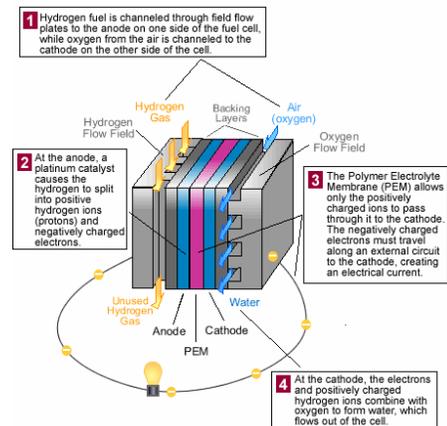
Net Energy?



What is a fuel cell?

A fuel cell is an electrochemical device that **combines hydrogen and oxygen to produce electricity, with water and heat as its by-product**. As long as fuel is supplied, the fuel cell will continue to generate power. Since the conversion of the fuel to energy takes place via an electrochemical process, not combustion, the process is clean, quiet and highly efficient – two to three times more efficient than fuel burning. No other energy generation technology offers the combination of benefits that fuel cells do. In addition to low or zero emissions, benefits include high efficiency and reliability, multi-fuel capability, siting flexibility, durability, and ease of maintenance. Fuel cells are also scalable and can be stacked until the desired power output is reached. Since fuel cells operate silently, they reduce noise pollution as well as air pollution and the waste heat from a fuel cell can be used to provide hot water or space heating for a home or office.

<http://www.fuelcells.org/>



<http://www.chevrolet.com/electriccar/>

<http://www.chevrolet.com/fuelcell/>



25



26

Wednesday, February 14, 2007 / Arizona Daily Star • A3

Smog linked to cancerous genetic changes in fetuses

WASHINGTON — Air pollution from traffic and power plants seems to cause genetic changes — the kind linked to cancer — in developing fetuses, a federally funded study released Tuesday has concluded. A three-month study of pregnant women in poor areas of New York City used hair locks to monitor the women's exposure to airborne carcinogens and then tested their babies' umbilical-cord blood after birth. Babies whose mothers were exposed to higher pollution levels had 50 percent more abnormalities in their chromosomes. Other studies have shown that these types of chromosomal changes increase the risk of cancer.

"This finding shows the process can begin as early as the month as a result of air pollution," said study author Professor Perera, the director of Columbia University's Center for Children's Environmental Health. "We know that these problems under their way before a child's first breath is even taken."

Perera's study didn't determine what parts of the babies' genes changed or if they all changed in the same way.

FRANK MUST DIE!

It's good riddance to Frank, so David is just fine, thank you

Health & Science

WASHINGTON — A brand of hot sauce who also happened to be named "Frank" — is no longer on the shelves of stores, after a study by the U.S. Food and Drug Administration (FDA) found that the sauce contained a chemical that causes cancer in laboratory animals.

"Frank is now dead and gone and never to be seen again," said FDA spokeswoman Jennifer M. Smith. "Frank the Frank sauce is no longer on the shelves of stores."

David's mother said she is not sure if she had the sauce on the other way, said Smith.

David's mother said she is not sure if she had the sauce on the other way, said Smith.

David's mother said she is not sure if she had the sauce on the other way, said Smith.



David's mother said she is not sure if she had the sauce on the other way, said Smith.

AZDStar, 16 Feb 2005



Perception of Risk

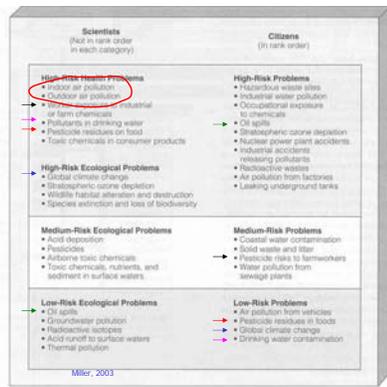
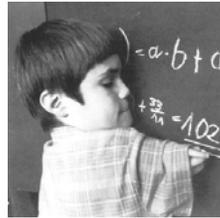


Figure 10-7 Comparative risk analysis of the most serious ecological and health problems according to scientists and citizens. The right side of the figure represents risks showing how U.S. citizens rate the ecological and health risks they perceive as the most serious. Why do you think there is such a perceptual difference? Do you think that the risks are really that different?

Thalidomide



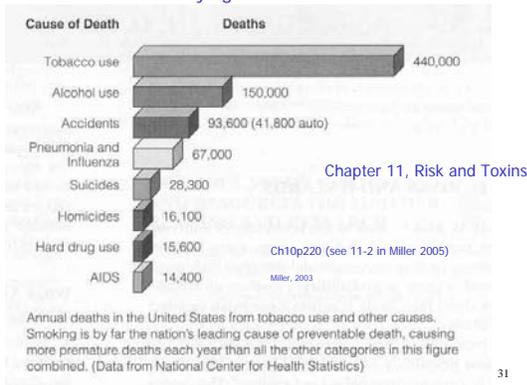
Thalidomide is a drug that was introduced on to the market on October 1, 1957 in West Germany. Thalidomide soon became a drug prescribed to pregnant women to combat symptoms associated with morning sickness. When taken during the first trimester of pregnancy, Thalidomide prevented the proper growth of the foetus resulting in horrific birth defects in thousands of children around the world. These children were born in the late 1950's and early 1960's and became known as "Thalidomide babies".

Of the 10,000 babies with "seal-like" limbs, only seventeen were born in the United States. The number was low because Dr. Frances Kelsey blocked the sale of the drug in this country.



Precautionary Principle

What are humans dying from?



31

What are humans dying from?

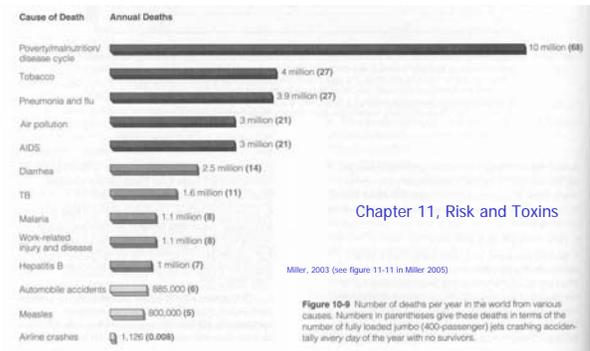
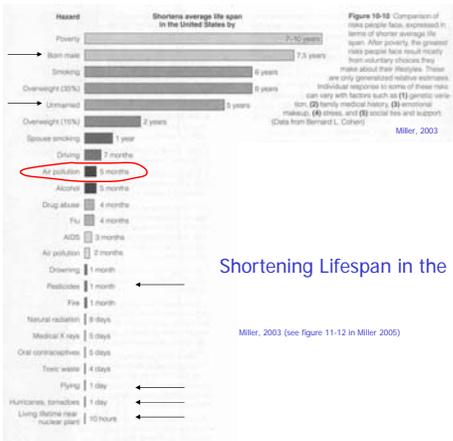


Figure 10-9 Number of deaths per year in the world from various causes. Numbers in parentheses give these deaths in terms of the number of fully loaded jumbo (400-passenger) jets crashing accidentally every day of the year with no survivors.



33

Ending Hunger and Disease (Miller p. 155):

- Immunizations
- Breast Feeding
- Sugar + Salt + Water to fight dehydration from diarrhea
- Vitamin A 2x/year to prevent blindness
- Family planning services
- Educate women
nutrition, sterilize drinking water, child care

Cost: \$5-10/child/year

34

Speth
Red Sky at Morning
Chapter 2, Lost in Eden

Biophilia (Keller&Wilson)

- Threats:
1. Land Use Conversion (1/3 forests gone, 1/2 wetlands)
 2. Land Degradation
 3. Freshwater Shortages
 4. Watercourse Modifications
 5. Invasive Species
 6. Overharvesting
 7. Climate Change
 8. Ozone Depletion
 9. Pollution

Only need ~\$30 billion/year to set aside 15% land area

35

Pollution

- 1-natural
- 2-anthropogenic
urban and industrial areas
industrial agriculture



- Sources:
- a. Point (smokestack, drainpipe, exhaust pipe)
 - b. Nonpoint (runoff with fertilizers and pesticides)

- Characteristics of Pollution:
1. Chemical nature
 2. Concentration
 3. Persistence
 - Degradable (human waste)
 - Persistent (DDT, plastics)
 - Nondegradable (lead, mercury)

36

- Air Pollution
- Primary
 - Secondary

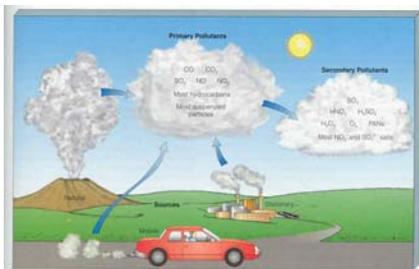


Figure 12-4 Sources and types of air pollutants. Human inputs of air pollutants may come from mobile sources (such as cars) and stationary sources (such as industrial and power plants). Some primary air pollutants react with one another or with other chemicals in the air to form secondary air pollutants.

Miller 11-22, 2003

See Miller 2005 Fig. 12-20

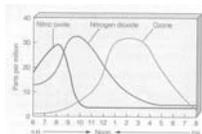
37

Air Pollution

Table 11-2 Major Classes of Air Pollutants	
Class	Examples
Carbon oxides	Carbon monoxide (CO), carbon dioxide (CO ₂)
Sulfur oxides	Sulfur dioxide (SO ₂), sulfur trioxide (SO ₃)
Nitrogen oxides	Nitric oxide (NO), nitrogen dioxide (NO ₂), nitrous oxide (N ₂ O) (NO and NO ₂ often are lumped together and labeled NO _x)
Volatile organic compounds (VOCs)	Methane (CH ₄), propane (C ₃ H ₈), chlorofluorocarbons (CFCs)
Suspended particulate matter (SPM)	Solid particles (dust, soot, asbestos, lead, nitrate and sulfate salts), liquid droplets (sulfuric acid, PCBs, dioxins, pesticides)
Photochemical oxidants	Ozone (O ₃), peroxyacetyl nitrates (PANs), hydrogen peroxide (H ₂ O ₂), aldehydes
Radioactive substances	Radon-222, iodine-131, strontium-90, plutonium-239
Hazardous air pollutants (HAPs), which cause health effects such as cancer, birth defects, and nervous system problems	Carbon tetrachloride (CCl ₄), methyl chloride (CH ₃ Cl), chloroform (CHCl ₃), benzene (C ₆ H ₆), ethylene bromide (C ₂ H ₄ Br ₂), formaldehyde (CH ₂ O)

Miller, 2003, see Miller 2005 Table 12-2

38



Miller, 2003
Figure 11-48 Typical daily pattern of changes in concentrations of air pollutants that lead to development of photochemical smog in a city such as Los Angeles, California.

Photochemical Smog

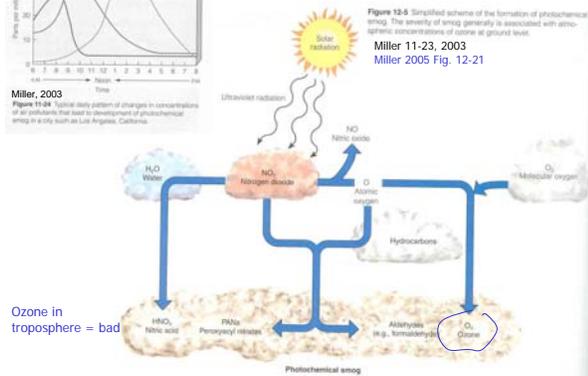


Figure 12-4 Simplified scheme of the formation of photochemical smog. The severity of smog generally is associated with atmospheric concentrations of ozone at ground level.

Miller 11-23, 2003
Miller 2005 Fig. 12-21

Ozone in troposphere = bad

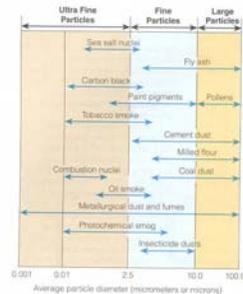
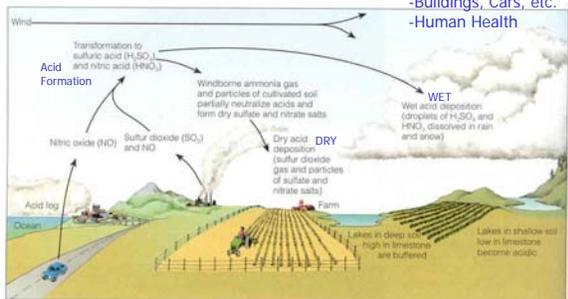


Figure 12-4 Suspended particulate matter consists of particles of solid matter and droplets of liquid that are small and light enough to remain suspended in the atmosphere for short periods (large particles) to long periods (small particles). Suspended particles are found in a wide variety of types and sizes, ranging in diameter from 0.001 micrometer to 100 micrometers (a micrometer, or micron, is one millionth of a meter, or about 0.0004 inches). Since 1987, the EPA has focused on the particles smaller than 10 microns (known as PM-10). In 1997, the agency began focusing on reducing emissions of ultrafine particles with diameters less than 2.5 microns (known as PM-2.5) because these particles are small enough to reach the lower part of human lungs and contribute to respiratory diseases.

40

Acid Deposition



Miller 11-26, 2003, Miller 2005 Figure 12-22
Figure 12-4 Acid deposition, which consists of rain, snow, dust, or gas with a pH lower than 5.6, is commonly called acid rain. Soils and lakes vary in their ability to buffer or remove excess acidity.

- Soil
- Water
- Fish
- Buildings, Cars, etc.
- Human Health

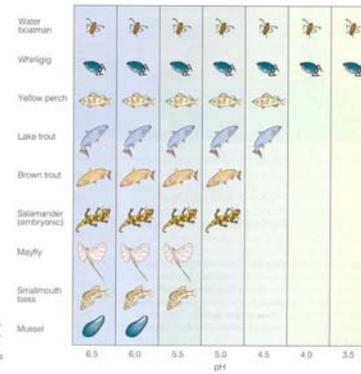


Figure 12-10 Fish and other aquatic organisms vary in their sensitivity to acidity. The figure shows the lowest pH (highest acidity) at which the various species can survive. Note that the greatest effects occur when the pH drops below 5.5.

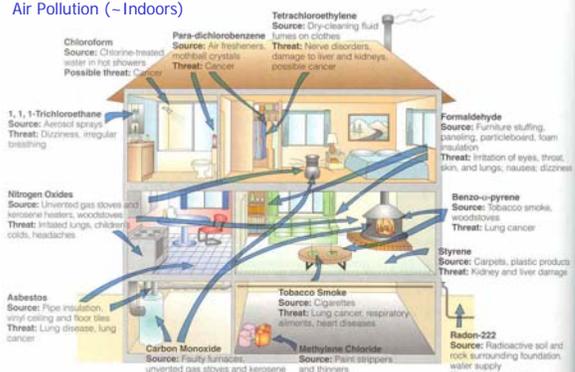
42



Figure 12-24 Solutions: methods for reducing acid deposition and its damage. Miller, 2005

43

Air Pollution (-Indoors)



See Miller 12-25, 2005

Figure 12-13 Some important indoor air pollutants. (Data from U.S. Environmental Protection Agency)

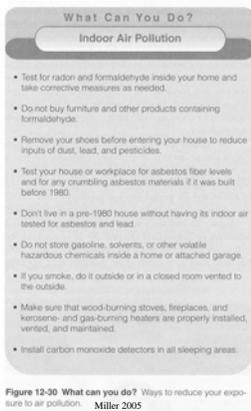


Figure 12-30 What can you do? Ways to reduce your exposure to air pollution. Miller 2005

45

Table 12-2 Major Outdoor Air Pollutants		
Carbon monoxide (CO) Description: Colorless, odorless gas that is poisonous to an breathing animals, some during the incomplete combustion of carbon-containing fuels (C + O ₂ → 2CO). Major human sources: Cigarette smoking (Case Study 2, 2005), incomplete burning of fossil fuels. About 7% (90% in cities) comes from motor vehicles. Health effects: Reacts with hemoglobin in red blood cells and reduces the ability of blood to transport oxygen. Above 100 ppm causes headaches, dizziness, and nausea. Can trigger heart attacks and angina. At high levels it causes collapse, coma, respiratory and heart failure, and death. Environmental effects: Reduces visibility and acid deposition of sulfur-containing fossil fuels such as coal and oil. Major human sources: Coal burning in power plants (90%) and industrial processes (10%). Health effects: Breathing problems for healthy people; restriction of always in people with asthma; chronic exposure can cause a permanent condition similar to bronchitis. According to the WHO, at least 825 million people are exposed to unsafe levels of sulfur dioxide from fossil fuel burning. Environmental effects: Reduces visibility and deposition of H ₂ SO ₄ can damage crops, soils, and aquatic life in lakes. Property damage: SO ₂ and H ₂ SO ₄ can corrode metals and	increase susceptibility to respiratory infections such as the flu and common colds; possible in young children and older adults. Environmental effects: Reduces visibility, acid deposition of H ₂ SO ₄ can damage trees, soil, and aquatic life in lakes. Property damage: H ₂ SO ₄ can corrode metals and eat away stone on buildings, statues, and monuments. NO _x can damage fabrics. Sulfur dioxide (SO₂) Description: Colorless, irritating fumes mostly from the combustion of sulfur-containing fossil fuels such as coal and oil (S + O ₂ → SO ₂). In the atmosphere can be converted to sulfuric acid (H ₂ SO ₄), a major component of acid deposition. Major human sources: Coal burning in power plants (90%) and industrial processes (10%). Health effects: Breathing problems for healthy people; restriction of always in people with asthma; chronic exposure can cause a permanent condition similar to bronchitis. According to the WHO, at least 825 million people are exposed to unsafe levels of sulfur dioxide from fossil fuel burning. Environmental effects: Reduces visibility and deposition of H ₂ SO ₄ can damage crops, soils, and aquatic life in lakes. Property damage: SO ₂ and H ₂ SO ₄ can corrode metals and	photochemical smog (Fig. 12-11). Major human sources: Chemical reaction with volatile organic compounds (VOCs), emitted mostly by cars and industrial and consumer products to form photochemical smog (Fig. 12-17). Health effects: Breathing problems, coughing, eye, nose, and throat irritation; aggravates chronic diseases, such as asthma, bronchitis, emphysema, and heart disease; reduce resistance to colds and pneumonia; they speed up lung tissue aging. Environmental effects: Ozone can damage plants and trees; smog can reduce visibility. Property damage: Can harm rubber, fabrics, and paints. Lead Description: Solid, toxic metal and its compounds, emitted into the atmosphere as particulate matter. Major human sources: Paint (old houses), smelters (metal refineries), lead manufacturing (storage batteries, leaded gasoline being phased out in most countries). Health effects: Accumulates in the body; brain and other nervous system damage and mental retardation (especially in children); digestive and other health problems; some neurodegenerative diseases; cause cancer in liver, stomach, and lung. Environmental effects: Can harm wildlife.
Nitrogen dioxide (NO₂) Description: Reddish-brown, irritating gas that gives photochemical smog its brownish color. In the atmosphere can be converted to nitric acid (HNO ₃), a major component of acid rain. Major human sources: Fossil fuel burning in motor vehicles (40%) and power and industrial plants (60%). Health effects: Lung irritation and damage; aggravates asthma and chronic bronchitis.	Environmental effects: Reduces visibility and acid deposition of H ₂ SO ₄ can damage crops, soils, and aquatic life in lakes. Property damage: SO ₂ and H ₂ SO ₄ can corrode metals and	Major human sources: Burning of fossil fuels (large quantities being generated and other fuels in vehicles (70%); agriculture (plowing, burning of fields, unpaid muck contribution). Health effects: Nose and eye irritation; lung damage and bronchitis; aggravates asthma and other respiratory ailments; toxic particulates irritate the lungs; particulates irritate the eyes, nose, and throat. Environmental effects: Ozone can damage plants and trees; smog can reduce visibility. Property damage: Can harm rubber, fabrics, and paints.
Formaldehyde Description: Colorless, pungent gas that is poisonous to an breathing animals, some during the incomplete combustion of carbon-containing fuels (C + O ₂ → 2CO). Major human sources: Cigarette smoking (Case Study 2, 2005), incomplete burning of fossil fuels. About 7% (90% in cities) comes from motor vehicles. Health effects: Reacts with hemoglobin in red blood cells and reduces the ability of blood to transport oxygen. Above 100 ppm causes headaches, dizziness, and nausea. Can trigger heart attacks and angina. At high levels it causes collapse, coma, respiratory and heart failure, and death. Environmental effects: Reduces visibility and acid deposition of sulfur-containing fossil fuels such as coal and oil. Major human sources: Coal burning in power plants (90%) and industrial processes (10%). Health effects: Breathing problems for healthy people; restriction of always in people with asthma; chronic exposure can cause a permanent condition similar to bronchitis. According to the WHO, at least 825 million people are exposed to unsafe levels of sulfur dioxide from fossil fuel burning. Environmental effects: Reduces visibility and deposition of H ₂ SO ₄ can damage crops, soils, and aquatic life in lakes. Property damage: SO ₂ and H ₂ SO ₄ can corrode metals and	Major human sources: Chemical reaction with volatile organic compounds (VOCs), emitted mostly by cars and industrial and consumer products to form photochemical smog (Fig. 12-17). Health effects: Breathing problems, coughing, eye, nose, and throat irritation; aggravates chronic diseases, such as asthma, bronchitis, emphysema, and heart disease; reduce resistance to colds and pneumonia; they speed up lung tissue aging. Environmental effects: Ozone can damage plants and trees; smog can reduce visibility. Property damage: Can harm rubber, fabrics, and paints. Lead Description: Solid, toxic metal and its compounds, emitted into the atmosphere as particulate matter. Major human sources: Paint (old houses), smelters (metal refineries), lead manufacturing (storage batteries, leaded gasoline being phased out in most countries). Health effects: Accumulates in the body; brain and other nervous system damage and mental retardation (especially in children); digestive and other health problems; some neurodegenerative diseases; cause cancer in liver, stomach, and lung. Environmental effects: Can harm wildlife.	Major human sources: Burning of fossil fuels (large quantities being generated and other fuels in vehicles (70%); agriculture (plowing, burning of fields, unpaid muck contribution). Health effects: Nose and eye irritation; lung damage and bronchitis; aggravates asthma and other respiratory ailments; toxic particulates irritate the lungs; particulates irritate the eyes, nose, and throat. Environmental effects: Ozone can damage plants and trees; smog can reduce visibility. Property damage: Can harm rubber, fabrics, and paints.
Styrene Description: Colorless, pungent gas that is poisonous to an breathing animals, some during the incomplete combustion of carbon-containing fuels (C + O ₂ → 2CO). Major human sources: Cigarette smoking (Case Study 2, 2005), incomplete burning of fossil fuels. About 7% (90% in cities) comes from motor vehicles. Health effects: Reacts with hemoglobin in red blood cells and reduces the ability of blood to transport oxygen. Above 100 ppm causes headaches, dizziness, and nausea. Can trigger heart attacks and angina. At high levels it causes collapse, coma, respiratory and heart failure, and death. Environmental effects: Reduces visibility and acid deposition of sulfur-containing fossil fuels such as coal and oil. Major human sources: Coal burning in power plants (90%) and industrial processes (10%). Health effects: Breathing problems for healthy people; restriction of always in people with asthma; chronic exposure can cause a permanent condition similar to bronchitis. According to the WHO, at least 825 million people are exposed to unsafe levels of sulfur dioxide from fossil fuel burning. Environmental effects: Reduces visibility and deposition of H ₂ SO ₄ can damage crops, soils, and aquatic life in lakes. Property damage: SO ₂ and H ₂ SO ₄ can corrode metals and	Major human sources: Burning of fossil fuels (large quantities being generated and other fuels in vehicles (70%); agriculture (plowing, burning of fields, unpaid muck contribution). Health effects: Nose and eye irritation; lung damage and bronchitis; aggravates asthma and other respiratory ailments; toxic particulates irritate the lungs; particulates irritate the eyes, nose, and throat. Environmental effects: Ozone can damage plants and trees; smog can reduce visibility. Property damage: Can harm rubber, fabrics, and paints.	Major human sources: Burning of fossil fuels (large quantities being generated and other fuels in vehicles (70%); agriculture (plowing, burning of fields, unpaid muck contribution). Health effects: Nose and eye irritation; lung damage and bronchitis; aggravates asthma and other respiratory ailments; toxic particulates irritate the lungs; particulates irritate the eyes, nose, and throat. Environmental effects: Ozone can damage plants and trees; smog can reduce visibility. Property damage: Can harm rubber, fabrics, and paints.
Carbon monoxide (CO) Description: Colorless, odorless gas that is poisonous to an breathing animals, some during the incomplete combustion of carbon-containing fuels (C + O ₂ → 2CO). Major human sources: Cigarette smoking (Case Study 2, 2005), incomplete burning of fossil fuels. About 7% (90% in cities) comes from motor vehicles. Health effects: Reacts with hemoglobin in red blood cells and reduces the ability of blood to transport oxygen. Above 100 ppm causes headaches, dizziness, and nausea. Can trigger heart attacks and angina. At high levels it causes collapse, coma, respiratory and heart failure, and death. Environmental effects: Reduces visibility and acid deposition of sulfur-containing fossil fuels such as coal and oil. Major human sources: Coal burning in power plants (90%) and industrial processes (10%). Health effects: Breathing problems for healthy people; restriction of always in people with asthma; chronic exposure can cause a permanent condition similar to bronchitis. According to the WHO, at least 825 million people are exposed to unsafe levels of sulfur dioxide from fossil fuel burning. Environmental effects: Reduces visibility and deposition of H ₂ SO ₄ can damage crops, soils, and aquatic life in lakes. Property damage: SO ₂ and H ₂ SO ₄ can corrode metals and	Major human sources: Burning of fossil fuels (large quantities being generated and other fuels in vehicles (70%); agriculture (plowing, burning of fields, unpaid muck contribution). Health effects: Nose and eye irritation; lung damage and bronchitis; aggravates asthma and other respiratory ailments; toxic particulates irritate the lungs; particulates irritate the eyes, nose, and throat. Environmental effects: Ozone can damage plants and trees; smog can reduce visibility. Property damage: Can harm rubber, fabrics, and paints.	Major human sources: Burning of fossil fuels (large quantities being generated and other fuels in vehicles (70%); agriculture (plowing, burning of fields, unpaid muck contribution). Health effects: Nose and eye irritation; lung damage and bronchitis; aggravates asthma and other respiratory ailments; toxic particulates irritate the lungs; particulates irritate the eyes, nose, and throat. Environmental effects: Ozone can damage plants and trees; smog can reduce visibility. Property damage: Can harm rubber, fabrics, and paints.
Formaldehyde Description: Colorless, pungent gas that is poisonous to an breathing animals, some during the incomplete combustion of carbon-containing fuels (C + O ₂ → 2CO). Major human sources: Cigarette smoking (Case Study 2, 2005), incomplete burning of fossil fuels. About 7% (90% in cities) comes from motor vehicles. Health effects: Reacts with hemoglobin in red blood cells and reduces the ability of blood to transport oxygen. Above 100 ppm causes headaches, dizziness, and nausea. Can trigger heart attacks and angina. At high levels it causes collapse, coma, respiratory and heart failure, and death. Environmental effects: Reduces visibility and acid deposition of sulfur-containing fossil fuels such as coal and oil. Major human sources: Coal burning in power plants (90%) and industrial processes (10%). Health effects: Breathing problems for healthy people; restriction of always in people with asthma; chronic exposure can cause a permanent condition similar to bronchitis. According to the WHO, at least 825 million people are exposed to unsafe levels of sulfur dioxide from fossil fuel burning. Environmental effects: Reduces visibility and deposition of H ₂ SO ₄ can damage crops, soils, and aquatic life in lakes. Property damage: SO ₂ and H ₂ SO ₄ can corrode metals and	Major human sources: Burning of fossil fuels (large quantities being generated and other fuels in vehicles (70%); agriculture (plowing, burning of fields, unpaid muck contribution). Health effects: Nose and eye irritation; lung damage and bronchitis; aggravates asthma and other respiratory ailments; toxic particulates irritate the lungs; particulates irritate the eyes, nose, and throat. Environmental effects: Ozone can damage plants and trees; smog can reduce visibility. Property damage: Can harm rubber, fabrics, and paints.	Major human sources: Burning of fossil fuels (large quantities being generated and other fuels in vehicles (70%); agriculture (plowing, burning of fields, unpaid muck contribution). Health effects: Nose and eye irritation; lung damage and bronchitis; aggravates asthma and other respiratory ailments; toxic particulates irritate the lungs; particulates irritate the eyes, nose, and throat. Environmental effects: Ozone can damage plants and trees; smog can reduce visibility. Property damage: Can harm rubber, fabrics, and paints.
Styrene Description: Colorless, pungent gas that is poisonous to an breathing animals, some during the incomplete combustion of carbon-containing fuels (C + O ₂ → 2CO). Major human sources: Cigarette smoking (Case Study 2, 2005), incomplete burning of fossil fuels. About 7% (90% in cities) comes from motor vehicles. Health effects: Reacts with hemoglobin in red blood cells and reduces the ability of blood to transport oxygen. Above 100 ppm causes headaches, dizziness, and nausea. Can trigger heart attacks and angina. At high levels it causes collapse, coma, respiratory and heart failure, and death. Environmental effects: Reduces visibility and acid deposition of sulfur-containing fossil fuels such as coal and oil. Major human sources: Coal burning in power plants (90%) and industrial processes (10%). Health effects: Breathing problems for healthy people; restriction of always in people with asthma; chronic exposure can cause a permanent condition similar to bronchitis. According to the WHO, at least 825 million people are exposed to unsafe levels of sulfur dioxide from fossil fuel burning. Environmental effects: Reduces visibility and deposition of H ₂ SO ₄ can damage crops, soils, and aquatic life in lakes. Property damage: SO ₂ and H ₂ SO ₄ can corrode metals and	Major human sources: Burning of fossil fuels (large quantities being generated and other fuels in vehicles (70%); agriculture (plowing, burning of fields, unpaid muck contribution). Health effects: Nose and eye irritation; lung damage and bronchitis; aggravates asthma and other respiratory ailments; toxic particulates irritate the lungs; particulates irritate the eyes, nose, and throat. Environmental effects: Ozone can damage plants and trees; smog can reduce visibility. Property damage: Can harm rubber, fabrics, and paints.	Major human sources: Burning of fossil fuels (large quantities being generated and other fuels in vehicles (70%); agriculture (plowing, burning of fields, unpaid muck contribution). Health effects: Nose and eye irritation; lung damage and bronchitis; aggravates asthma and other respiratory ailments; toxic particulates irritate the lungs; particulates irritate the eyes, nose, and throat. Environmental effects: Ozone can damage plants and trees; smog can reduce visibility. Property damage: Can harm rubber, fabrics, and paints.
Carbon monoxide (CO) Description: Colorless, odorless gas that is poisonous to an breathing animals, some during the incomplete combustion of carbon-containing fuels (C + O ₂ → 2CO). Major human sources: Cigarette smoking (Case Study 2, 2005), incomplete burning of fossil fuels. About 7% (90% in cities) comes from motor vehicles. Health effects: Reacts with hemoglobin in red blood cells and reduces the ability of blood to transport oxygen. Above 100 ppm causes headaches, dizziness, and nausea. Can trigger heart attacks and angina. At high levels it causes collapse, coma, respiratory and heart failure, and death. Environmental effects: Reduces visibility and acid deposition of sulfur-containing fossil fuels such as coal and oil. Major human sources: Coal burning in power plants (90%) and industrial processes (10%). Health effects: Breathing problems for healthy people; restriction of always in people with asthma; chronic exposure can cause a permanent condition similar to bronchitis. According to the WHO, at least 825 million people are exposed to unsafe levels of sulfur dioxide from fossil fuel burning. Environmental effects: Reduces visibility and deposition of H ₂ SO ₄ can damage crops, soils, and aquatic life in lakes. Property damage: SO ₂ and H ₂ SO ₄ can corrode metals and	Major human sources: Burning of fossil fuels (large quantities being generated and other fuels in vehicles (70%); agriculture (plowing, burning of fields, unpaid muck contribution). Health effects: Nose and eye irritation; lung damage and bronchitis; aggravates asthma and other respiratory ailments; toxic particulates irritate the lungs; particulates irritate the eyes, nose, and throat. Environmental effects: Ozone can damage plants and trees; smog can reduce visibility. Property damage: Can harm rubber, fabrics, and paints.	Major human sources: Burning of fossil fuels (large quantities being generated and other fuels in vehicles (70%); agriculture (plowing, burning of fields, unpaid muck contribution). Health effects: Nose and eye irritation; lung damage and bronchitis; aggravates asthma and other respiratory ailments; toxic particulates irritate the lungs; particulates irritate the eyes, nose, and throat. Environmental effects: Ozone can damage plants and trees; smog can reduce visibility. Property damage: Can harm rubber, fabrics, and paints.

Data from U.S. Environmental Protection Agency

Miller 2005

46

"Are you saving the industry a billion dollars but taking away \$10 billion worth of benefits for the general public?" Amar asked.

Mercury...

Mercury is a toxic metal emitted by industrial sources. U.S. power plants emit 48 tons a year, and the new rule establishes an emissions-trading program that is expected to lower emissions to about 31 tons by 2010 and to about 15 tons by 2026. The Harvard analysis was based on similar targets in President Bush's "Clear Skies" legislative proposal. In most cases, mercury toxicity results from eating fish: Industrial emissions fall from the air into water and are taken up by fish. Because the metal does not break down, it moves steadily up the food chain to species that people consume. A major reason for the dramatic difference in the health benefit estimates was that the EPA looked only at the effects of reducing mercury levels in freshwater fish, but most of the fish Americans eat comes from oceans.

"Some very large share of mercury exposure comes from tuna," Hammitt said. "And while it's true that our power plants have less effect on tuna than on [freshwater] northern pike, if you ignore the saltwater pathway you'll miss a lot of the benefit."

Even though U.S. power plants contribute only about 1 percent of the mercury in the oceans, reducing even that small amount makes a difference, he said. The EPA has said that ocean species such as tuna, pollock, shrimp and halibut account for two-thirds of the mercury Americans consume, while catfish, the largest source of mercury among freshwater fish, accounts for only 3 percent.

By Shankar Vedantam
Washington Post Staff Writer
Tuesday, March 22, 2005; Page A01

47

Illinois

Why is the statewide methylmercury advisory only for predator species?

Mercury stays in the environment for a long time. Small organisms absorb mercury from water and sediment; these organisms then are eaten by smaller fish. Predator fish eat the smaller fish and methylmercury is accumulated up the food chain. Larger fish have the highest amounts of methylmercury stored in their bodies. Predator species for Illinois include all species of black bass (largemouth, smallmouth and spotted), striped bass, white bass, hybrid bass, walleye, sauger, saugeye, flathead catfish, muskellunge and northern pike.

How does methylmercury get into bodies of water in Illinois?

Mercury is a metal that occurs naturally in small amounts in the environment. It also comes from burning coal or trash and from industry. Mercury gets into lakes and rivers in several ways, including rain and runoff. When conditions are right in the water, certain kinds of bacteria change metallic mercury into the more toxic methylmercury. Methylmercury is stored in the muscle of fish, the part of the fish people eat.

What are the potential health effects for people who eat fish contaminated with methylmercury?

The developing nervous systems of fetuses and children could be damaged if exposed to even small amounts of methylmercury. At high doses, methylmercury can affect the central nervous system (causing such health problems as memory loss and slurred speech) and can cause kidney damage and failure, and gastrointestinal damage. The possible health effects depend on how much methylmercury is stored in the fish and how much fish is eaten over a period of time. Based on the amounts of methylmercury detected in predator sport fish in Illinois, it is unlikely that people would experience adverse health effects associated with exposure to high doses.

48