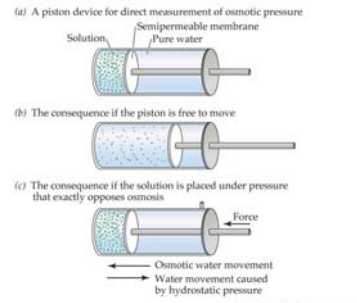


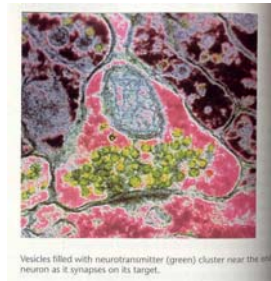
Lecture 5, 28 Jan 2008

Vertebrate Physiology
ECOL 437 (MCB/VetSci 437)
Univ. of Arizona, spring 2008

Kevin Bonine & Kevin Oh



1. Water, Solutes, Osmosis (Ch 3)
2. Intro Nervous System Fxn (Ch10)



http://eebweb.arizona.edu/eeb_course_websites.htm

Housekeeping, 28 January 2008



Upcoming Readings

today: [Ch 3&10](#)

Wed 30 Jan: [Ch 10&11](#)

LAB Wed 30 Jan: [Bisbal & Specker](#), plus two optional papers
(see website for links to papers; "worksheet" via email)

Fri 01 Feb: [Ch11](#)

Lab discussion leaders: [30 Jan](#)

1pm – [Josh, Seth](#)

3pm – [Aaron, Adam](#)

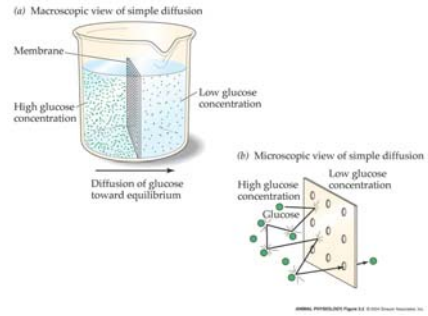
Lab discussion leaders: [06 Feb](#)

1pm – [Rittner, Whitney](#)

3pm – [Roxanne, Maria](#)

2

Vertebrate Physiology 437

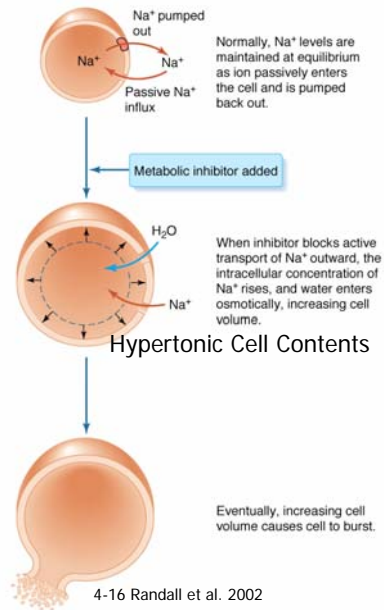
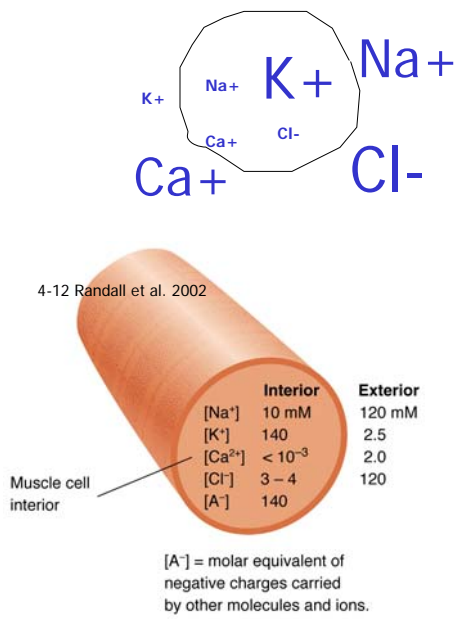


Chapter 3

Movement of Solutes and Water

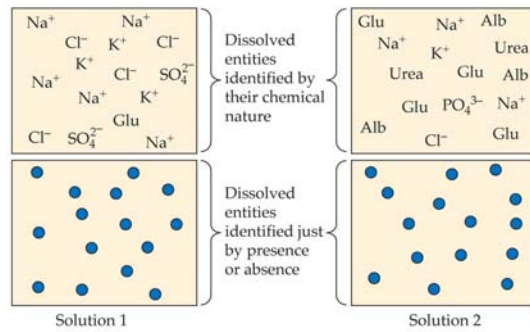
3

Osmotic Properties of Cells and Relative Ion Concentrations



Colligative Properties

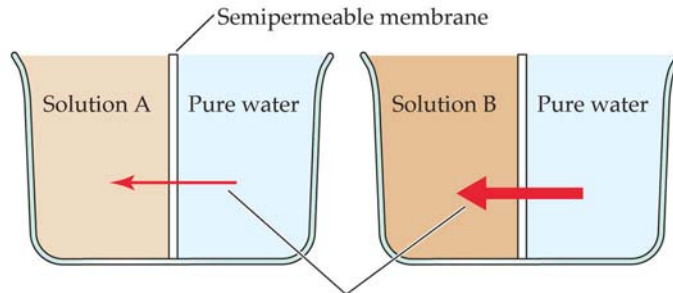
- Osmotic Pressure
- Freezing Point
- Water Vapor Pressure (boiling point; evaporation)



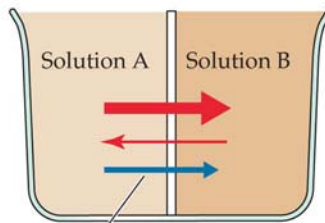
Hill et al 2004

ANIMAL PHYSIOLOGY, Figure 3.16 © 2004 Sinauer Associates, Inc.

5



Measurements on two solutions separated from pure water



Osmosis when the two solutions are separated from each other

Hill et al 2004

ANIMAL PHYSIOLOGY, Figure 3.17 © 2004 Sinauer Associates, Inc.

$$6 \times 10^{23}$$

Osmolarity

1 osmolar solution (Osm)

has 1 Avogadro's number of dissolved particles/liter solvent

1 milliosmolar solution (mOsm)

has 0.001 Avogadro's number of dissolved particles/liter solvent

7

What osmolarity do you get if you add 6×10^{23} molecules of [glucose](#) to a liter of water?

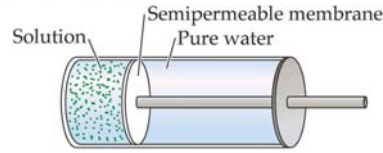
What osmolarity do you get if you add 6×10^{23} molecules of table [salt](#) to a liter of water?

NaCl (strong electrolyte)

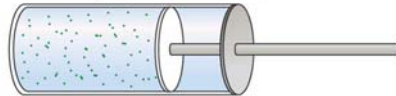
8

Osmotic Pressure
Vs.
Hydrostatic Pressure

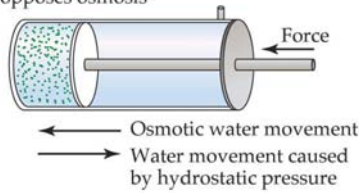
(a) A piston device for direct measurement of osmotic pressure



(b) The consequence if the piston is free to move



(c) The consequence if the solution is placed under pressure that exactly opposes osmosis



Hill et al 2004

ANIMAL PHYSIOLOGY, Figure 3.16 © 2004 Sinauer Associates, Inc.

9

Difference in osmotic potential

$$\text{Rate of Osmosis} = K \frac{\Pi_1 - \Pi_2}{X}$$

Proportionality
Coefficient
(~ permeability
and temp)

Distance
between
solutions

10

Electrochemical equilibrium

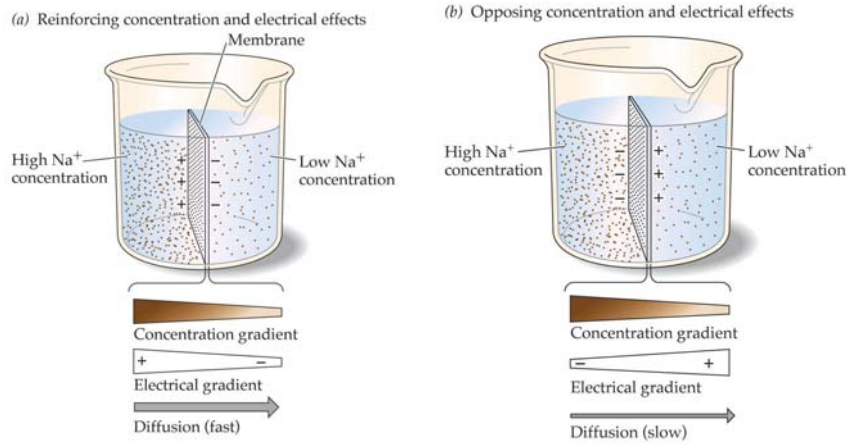


Fig 3.6, Hill et al 2004

11

Movement Across Membranes

Electrochemical Gradient

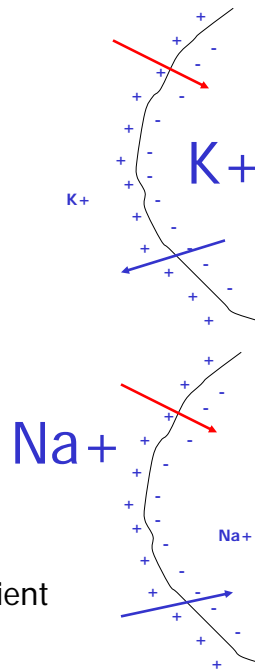
Electrical gradient

Concentration gradient

Electrochemical equilibrium

Equilibrium potential (E_x in mV)

when [X] gradient = electrical gradient

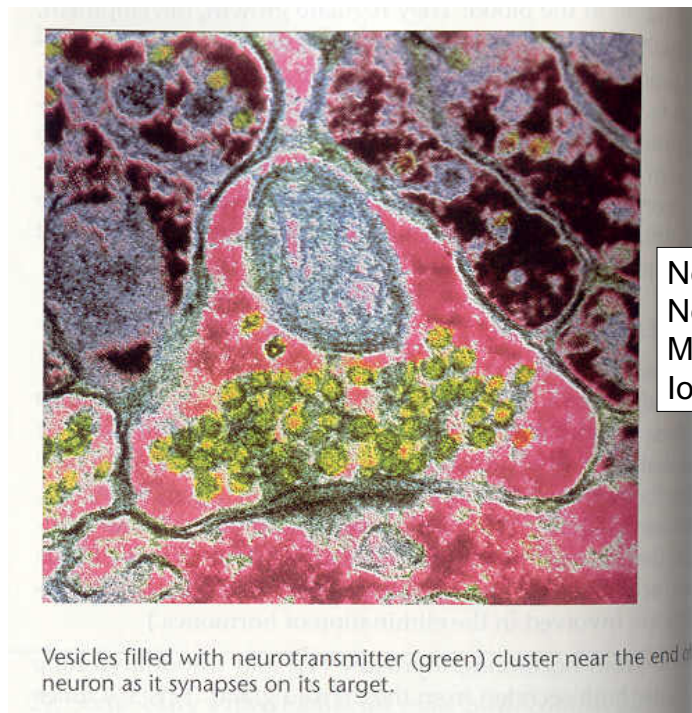


12

Equilibrium potential (E_x in mV)

“Every ion’s goal in life is to make the membrane potential equal its own equilibrium potential (E_x in mV)”

13



p. 214, Silverthorn
2001, 2nd ed.
Human Physiology,
Prentice Hall

Nervous System
Neurons
Membranes
Ions

14

Nervous System

Comprises

- Neurons / Nerve Cells
- Glial Cells (support)

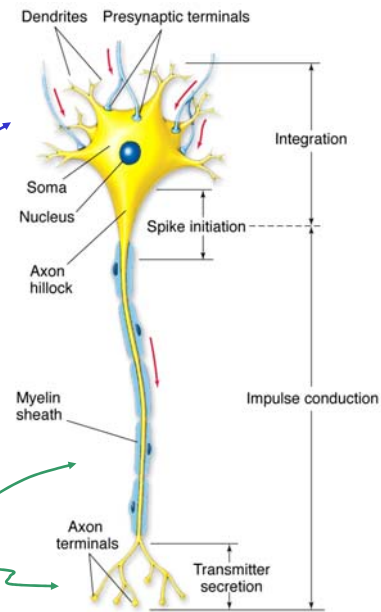
- Signalling via combination of **Electrical and Chemical**

- Integrate information

AFFERENT

- Coordinate Response

EFFERENT



15
5-2 Randall et al. 2002

Organization of the Nervous System

Three main functions:

1. **Sensory Reception** (converts environmental stimulus to elect/chem)
2. **Central Processing**
3. **Motor Output**

Divided into CNS and PNS

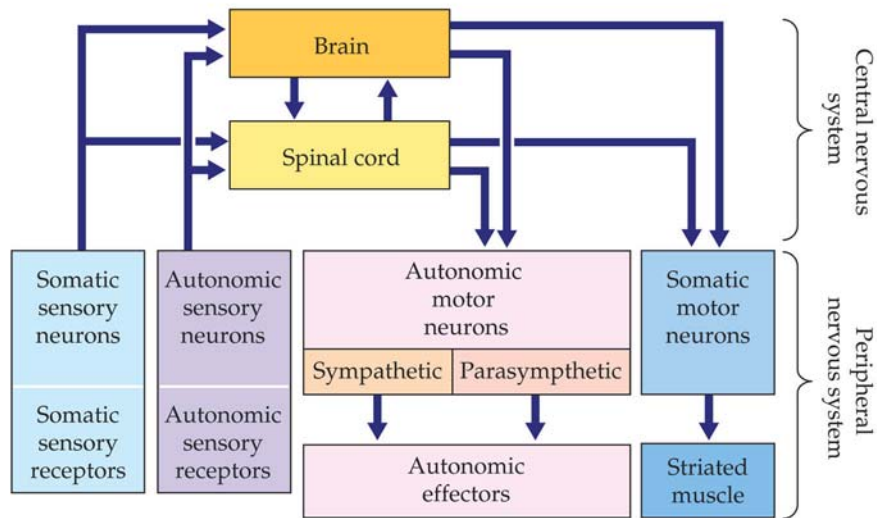
A. CNS = **Central Nervous System**

- Brain and Spinal Cord
(and eyes and interneurons)

B. PNS = **Peripheral Nervous System**

- most sensory and motor axons

16

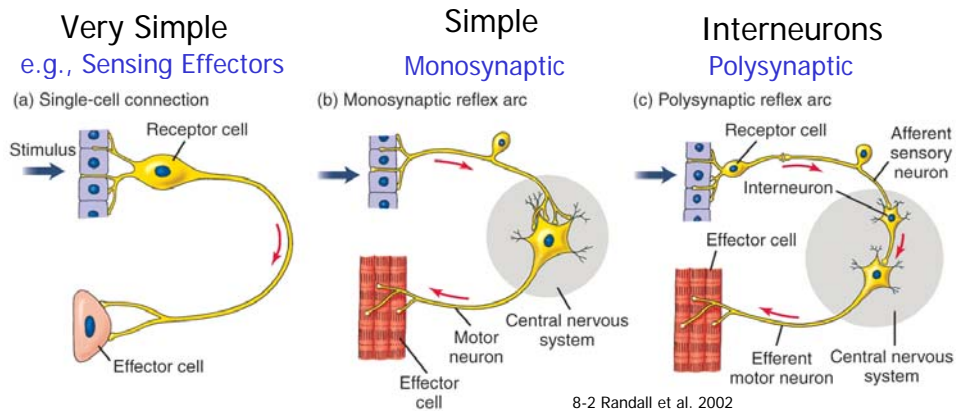


Hill et al. 2004, Fig 10.7

ANIMAL PHYSIOLOGY, Figure 10.7 © 2004 Sinauer Associates, Inc.

Flow of Information

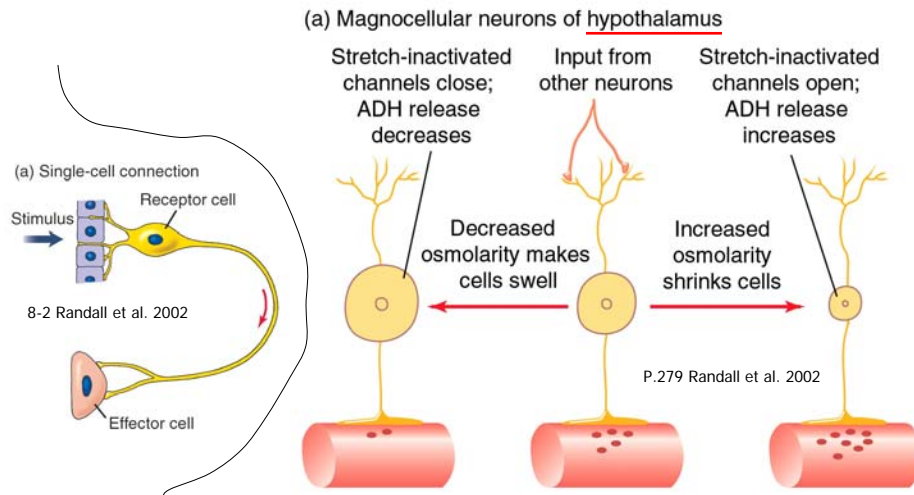
Afferent Signal -> CNS -> Efferent Signal -> Response



Sensing Effectors

Sensors with **Afferent and Efferent Properties/Homeostasis**

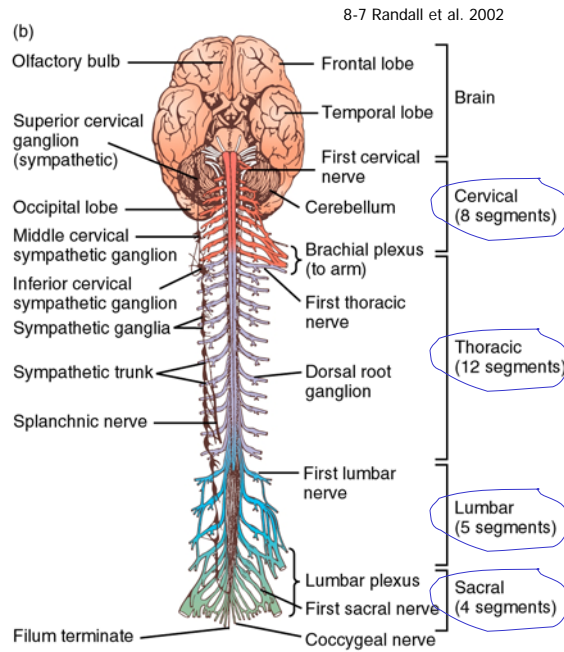
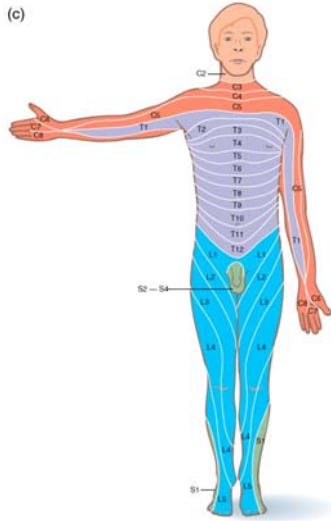
e.g., osmolarity and antidiuretic hormone (ADH)



Evolution of Nervous System:

- Based on the neuron
- Elaboration of Reflex Arc
- Group neurons into CNS
- More neurons in complex organisms
- New structures added on to old (not replaced)
- Size of CNS region correlated with importance
- Topological Maps

CNS - Spinal Cord
 - Retains evidence of segmented ancestors



CNS - Spinal Cord
Functional Anatomy

White matter = myelin
 Gray matter = somata and dendrites

Cerebrospinal Fluid in spinal canal

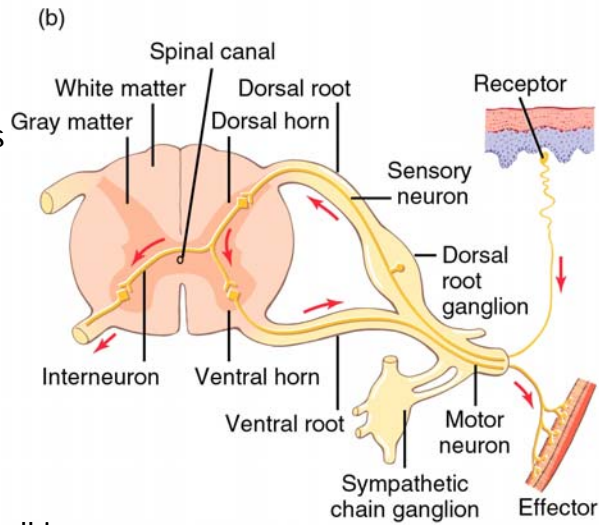
Dorsal root, horn = ~afferent

Ventral root, horn = ~efferent

Spinal Reflexes:

- locomotion / walking
- chicken w/o head

Dorsal Root Ganglion (PNS)
 - afferent sensory somata

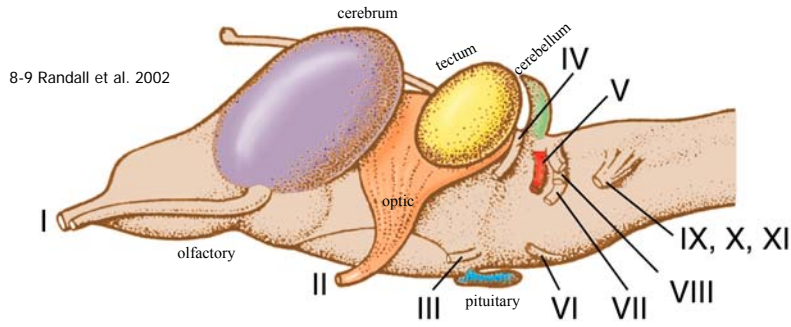


8-8 Randall et al. 2002

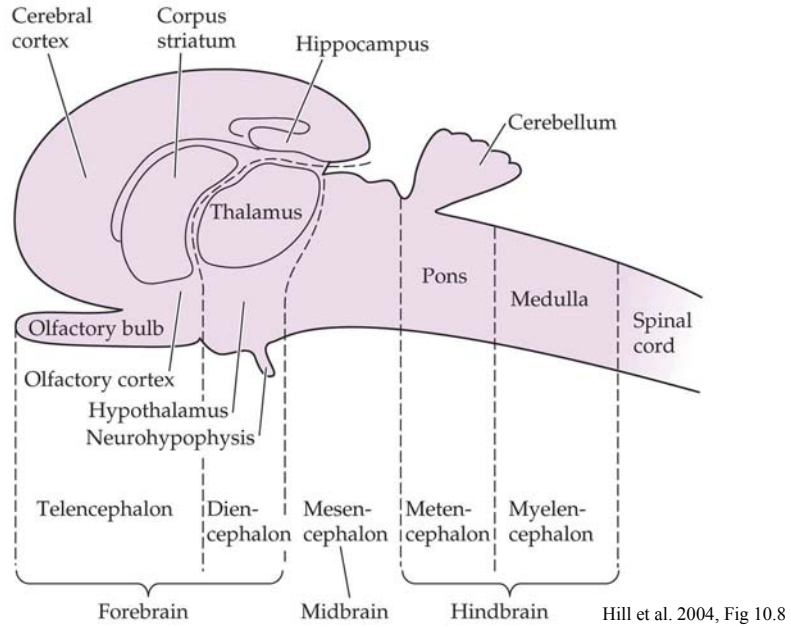
22

CNS - Brain

Vertebrate **bilaterally paired nerve** connections to periphery



- | | | | |
|---|---|------------------------|--|
| <p>Sensory</p> <ul style="list-style-type: none"> I Olfactory II Optic III Oculomotor IV Trochlear | <ul style="list-style-type: none"> V Trigeminal VI Abducens VII Facial VIII Vestibulocochlear | <p>← Both →</p> | <ul style="list-style-type: none"> IX Glossopharyngeal X Vagus XI Spinal accessory XII Hypoglossal <p>Motor</p> |
|---|---|------------------------|--|



Hill et al. 2004, Fig 10.8
ANIMAL PHYSIOLOGY, Figure 10.8 © 2004 Sinauer Associates, Inc.

CNS - Brain Functional Anatomy

-Medulla oblongata

Respiration, autonomic funct, some sensory (hearing, equil.)

-Cerebellum

Coordinate motor output

Integrates info. from proprioceptors (stretch and joint)
visual, auditory

More convoluted (↑ s.a.) in higher groups

Birds with large cerebellum to handle 3D flight

-Pons (and tectum)

Integrate and communicate

Visual, tactile, auditory maps

~ body movement coordination in some groups

-Cerebral Cortex

In higher groups takes over function of tectum

25

CNS - Brain Functional Anatomy (con't)

-Thalamus

Sensory and motor coordination

Often communicates with cerebral cortex

-Amygdala

Processes info. and output related to emotions

-Hypothalamus

Also involved in emotions

Body temp, eating, drinking, sex

Water and electrolyte balance

-Olfactory Bulb

Key sense in many vertebrate groups

Anterior position

-Cerebrum (covered by cerebral cortex)

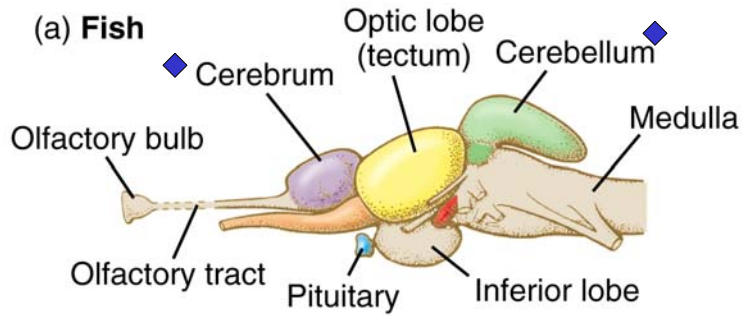
More evolved in higher groups (size and folds)...

Olfaction straight to
cerebrum w/o going
through thalamus
(in mammals)

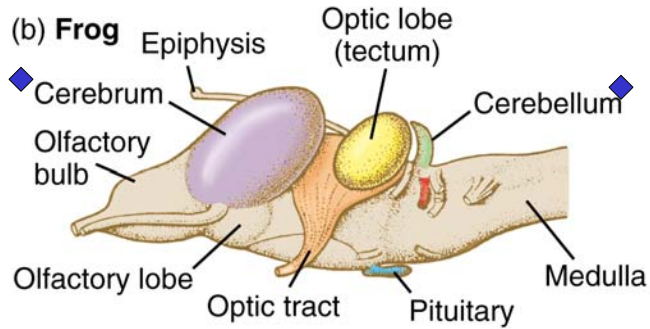
26

CNS - Brain

◆ Note change in size in diff. groups

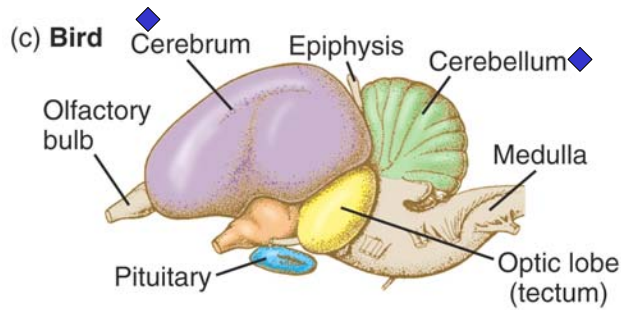


8-10 Randall et al. 2002

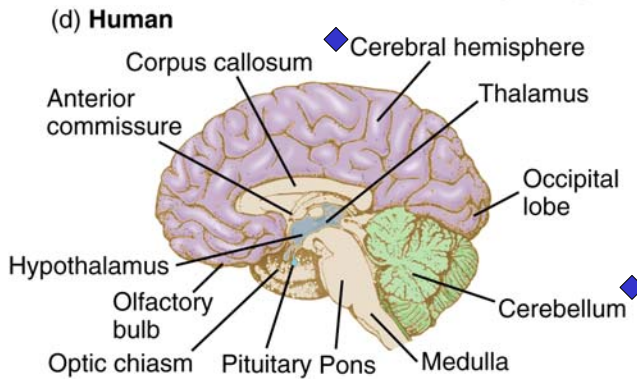


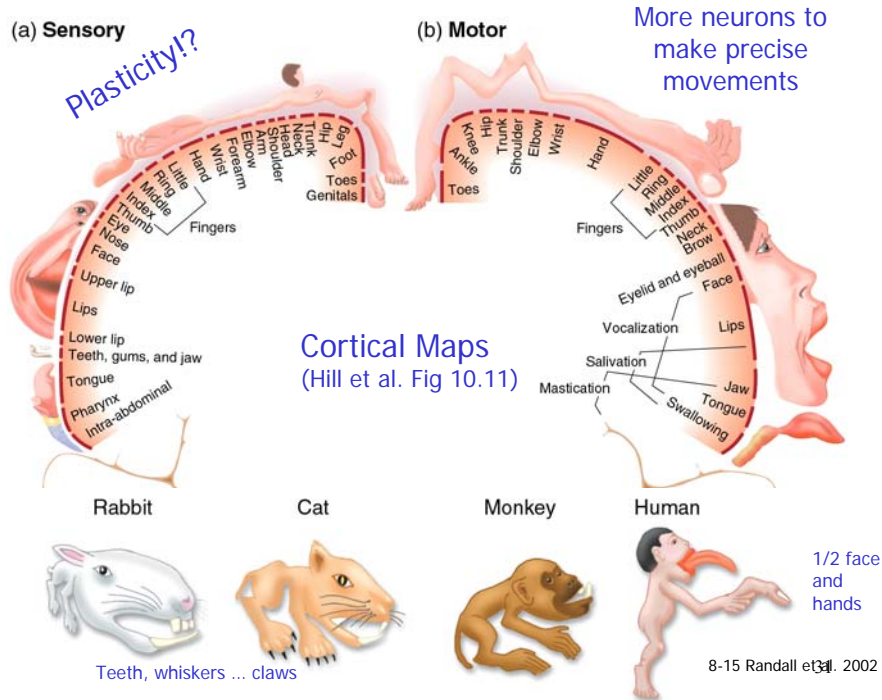
CNS - Brain

◆ Note change in size in diff. groups



8-10 Randall et al. 2002





Structural and Functional Regions

CNS

- Most neuronal somata
incl. motor neurons
- Interneurons

Nuclei = collections of somata w/ similar function

Tracts = bundles of axons from nuclei

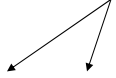
PNS

- Nervous system outside CNS

Nerves = axon bundles from sensory + motor neurons

Ganglia = somata of some autonomic neurons and most sensory neurons

Nerve usually with both Afferent and Efferent axons



Structural and Functional Regions

8-6 Randall et al. 2002

- Efferent NS

1. Somatic/Voluntary

- skeletal muscle

2. Autonomic

- smooth muscle
- cardiac muscle
- glands
- "housekeeping"

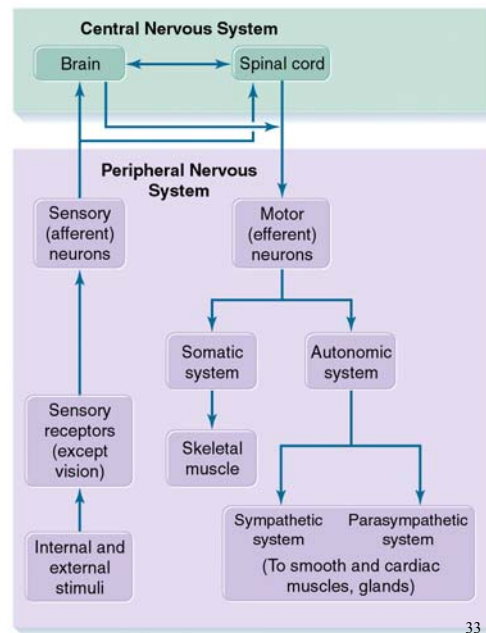
A. Sympathetic

~ fight or flight

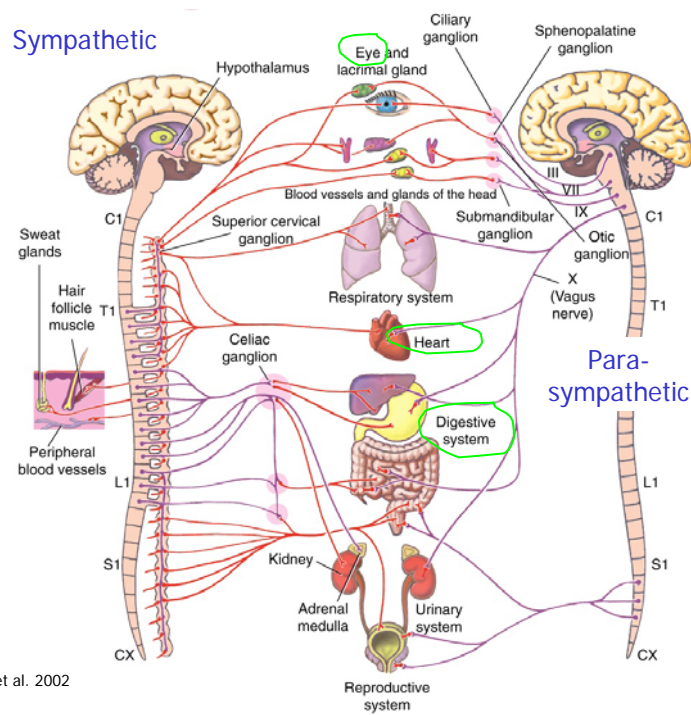
B. Parasympathetic

~ rest and digest

C. Enteric



Sympathetic



8-18 Randall et al. 2002

Autonomic NS (vs Voluntary/Somatic)

Antagonistic Groups in Balance:

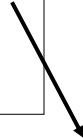
- A. Sympathetic (f or f)
- B. Parasympathetic (r + d)

Both function via **reflex arcs**, but often **opposite** effects

Efferent signal with two neurons:

- 1. **Preganglionic** (NT released is Acetylcholine [ACh])
- 2. **Postganglionic** (PNS, receptor is nicotinic ACh)

Difference between Symp. and Para. is in:

- 1. CNS origin
 - 2. Location of postganglionic somata
 - 3. Postganglionic NT
 - 4. Receptors on target tissues
- 

35

Autonomic NS

Sympathetic

- 2-Postganglionic somata **nearer CNS** in chain ganglia
- 3- Postganglionic NT is **Norepinephrine**
- 4-Effector receptor is **alpha or beta adrenergic**

Parasympathetic

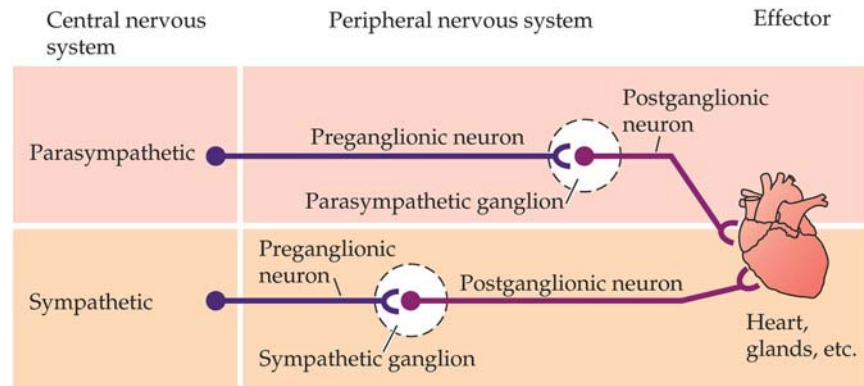
- 2-Postganglionic somata **near effector**, or in effector organ
- 3- Postganglionic NT is **ACh**
- 4-Effector receptor is **muscarinic ACh**

Difference between Symp. and Para. is in:

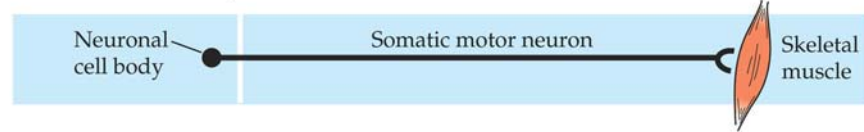
- 1. CNS origin
- 2. Location of postganglionic somata
- 3. Postganglionic NT
- 4. Receptors on target tissues

36

(a) Autonomic nervous system

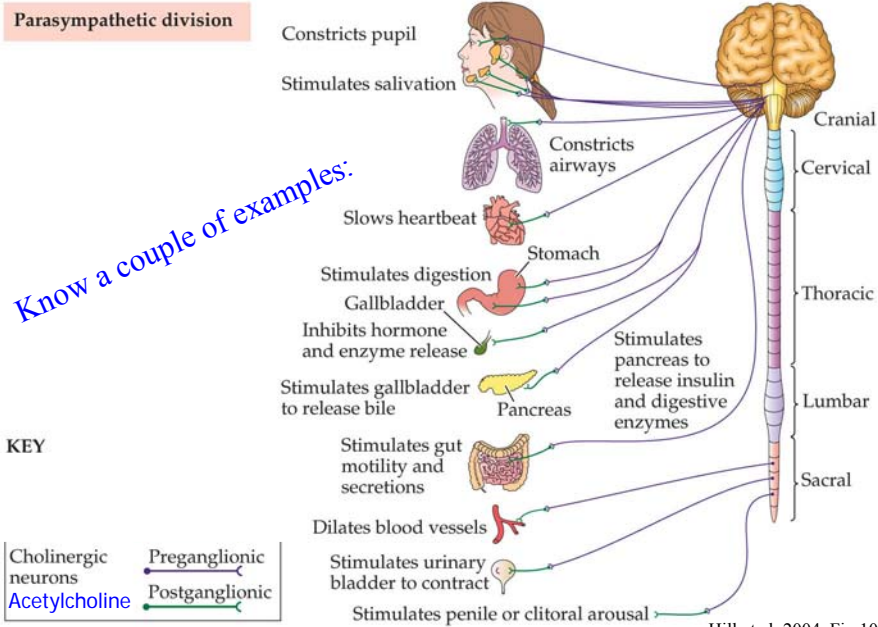


(b) Somatic nervous system



Hill et al. 2004, Fig 10.12

ANIMAL PHYSIOLOGY, Figure 10.12 © 2004 Sinauer Associates, Inc.



Hill et al. 2004, Fig 10.13

ANIMAL PHYSIOLOGY, Figure 10.13 (Part 2) © 2004 Sinauer Associates, Inc.

Sympathetic division

Know a couple of examples:

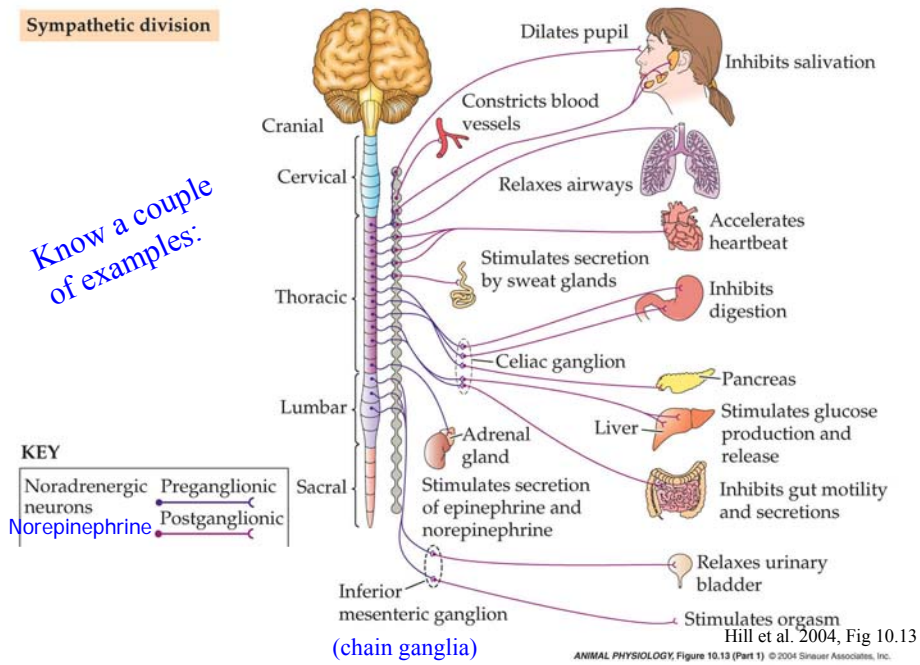


TABLE 10.2 Major actions of sympathetic and parasympathetic divisions in vertebrates

Process	Parasympathetic effect	Sympathetic effect
Digestion: gastrointestinal secretion and motility	Stimulates	Inhibits
Heartbeat	Slows	Increases rate and force
Blood vessels	Usually dilates	Constricts vessels to kidneys and gut; dilates vessels to skeletal muscles
Blood pressure	Decreases	Increases
Lung passages	Constricts	Dilates
Secretion of epinephrine and norepinephrine by adrenal medullary glands	—	Stimulates

Hill et al. 2004

ANIMAL PHYSIOLOGY, Table 10.2 © Sinauer Associates, Inc.

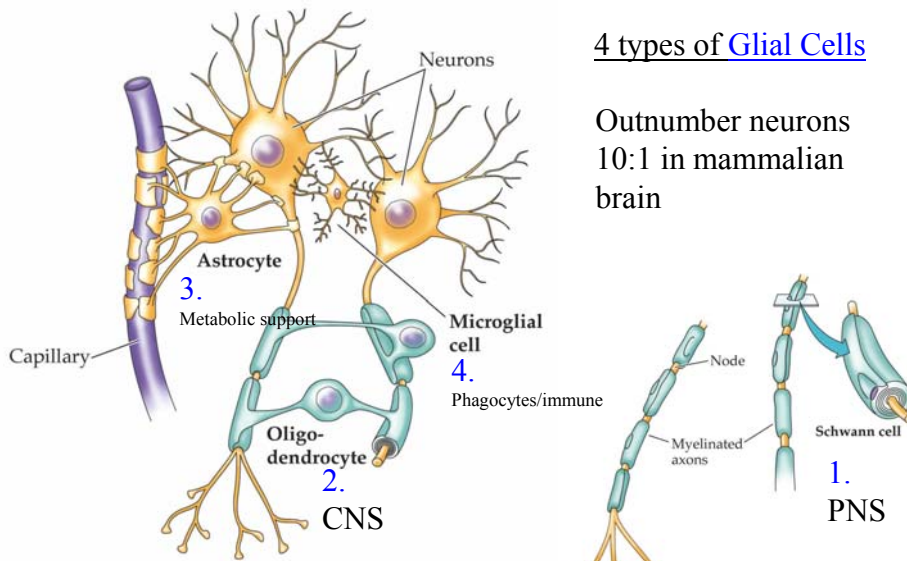


ANIMAL PHYSIOLOGY, Ch. 11 OpenStax © 2016 Sinauer Associates, Inc.

“Squid axons are important to physiologists, and to the squid.”
 Hill et al. 2004, p.281

Sir Alan Hodgkin, Nobel Prize 1963

41



4 types of Glial Cells

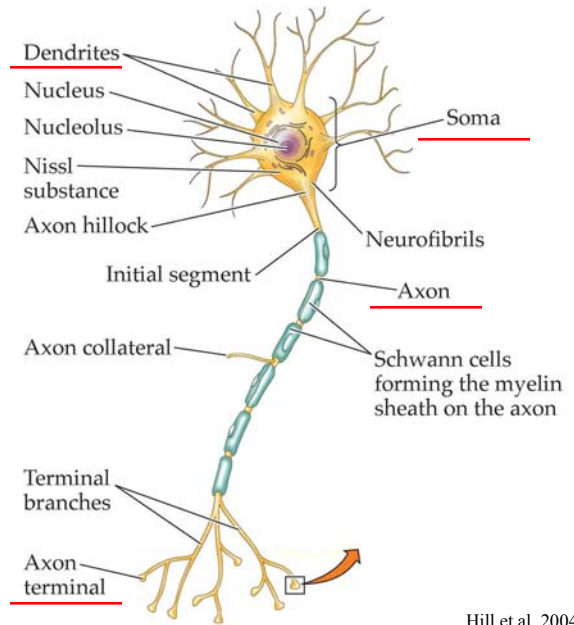
Outnumber neurons
 10:1 in mammalian
 brain

Glial Cells
 Support Function of NS

Hill et al. 2004, Fig 11.2

42

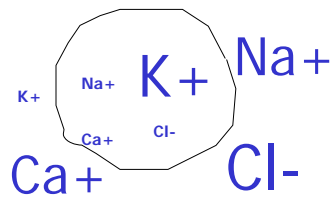
Neurons:



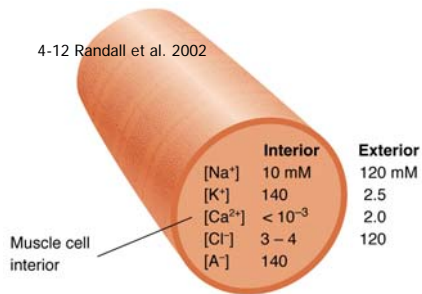
Hill et al. 2004, Fig 11.1

ANIMAL PHYSIOLOGY, Figure 11.1 (Part 1) © 2004 Sinauer Associates, Inc.

Osmotic Properties of Cells and Relative Ion Concentrations



4-12 Randall et al. 2002



[A⁻] = molar equivalent of negative charges carried by other molecules and ions.

To understand how the NS works we need to return to Membrane Details

Movement Across Membranes

Electrochemical Gradient

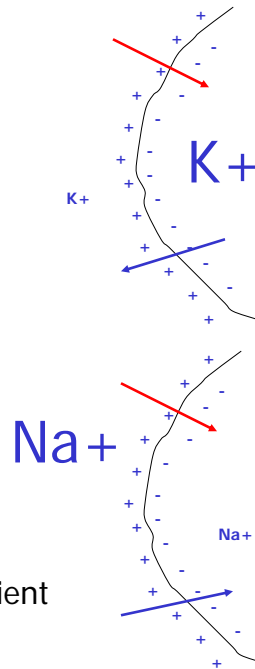
Electrical gradient

Concentration gradient

Electrochemical
equilibrium

Equilibrium potential (E_x in mV)

when $[X]$ gradient = electrical gradient



45

Equilibrium potential (E_x in mV)

“Every ion’s goal in life is to make the membrane potential equal its own equilibrium potential (E_x in mV)”

46

Membrane Potential

1. To change V_m :

A Small Number of Ions actually move relative to the number present both inside and outside the cell

2. Concentration gradients...

(previously established by ATPase pumps) are **not abolished** when the channels for an ion species open

[Gradients allow for 'work' to be done, e.g., action potential sends signal along axon]

47

Membrane Potential

3. Driven by ions that are permeable to the membrane (and have different $[]_{in}$ as compared to $[]_{out}$ a.k.a. gradient created with ATP)

- K^+ for example

4. Equilibrium Potential (E_x in mV):

~The equilibrium potentials of all the permeable ions (a function of their established gradients) will determine the membrane potential of a cell

5. emf determines which direction a given ion (X) will move when the membrane potential is known

$$emf_x = V_m - E_x$$

48

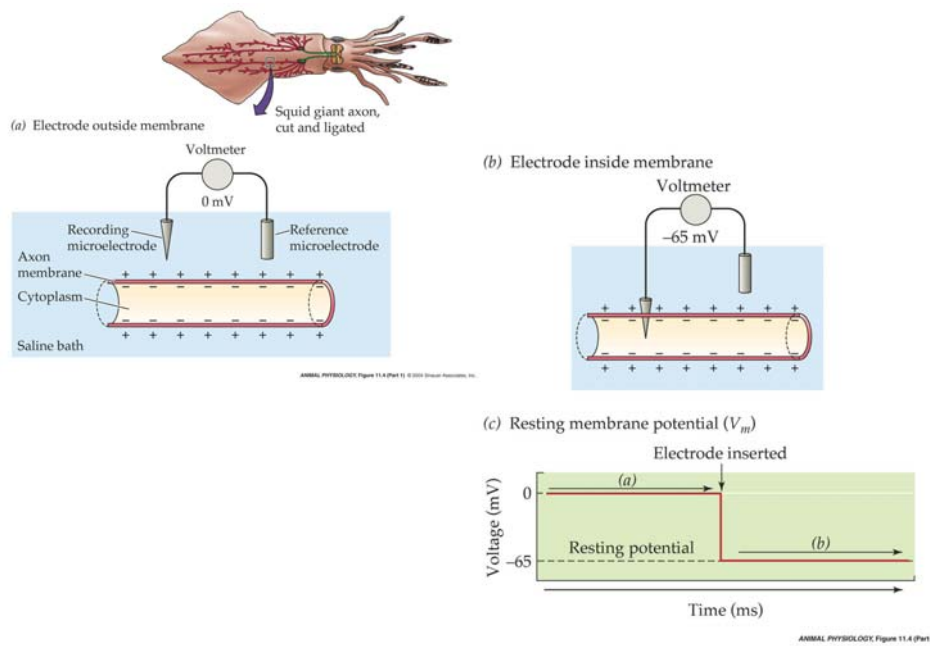
Membrane Potential

6. Resting Membrane Potential driven by K^+ efflux and, to a lesser extent, Na^+ influx

7. Na^+/K^+ ATPase pump generates gradients that, for these permeable ions, determine membrane potential



49



Hill et al. 2004, Fig 11.4

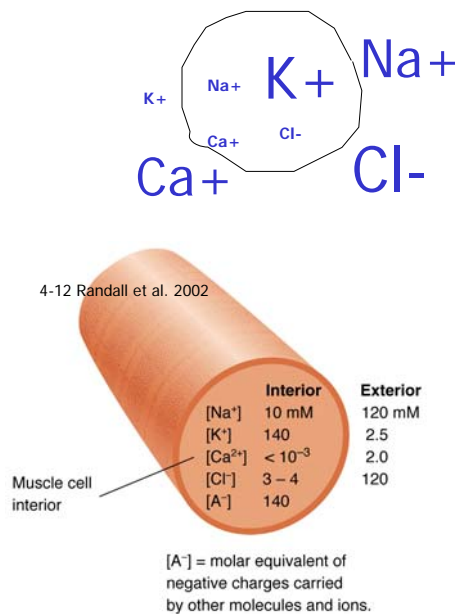
50

Table 5-1 Examples of ion channels found in axons

Randall et al. 2002

Channel	Current through channel	Characteristics	Selected blockers	Function
Leak channel (open in resting axon)	I_K (leak)	Produces relatively high P_K of resting cell	Partially blocked by tetraethylammonium (TEA)	Largely responsible for V_{rest}
Voltage-gated Na^+ channel	I_{Na}	Rapidly activated by depolarization; becomes inactivated even if V_m remains depolarized	Tetrodotoxin (TTX)	Produces rising phase of AP
Voltage-gated Ca^{2+} channel	I_{Ca}	Activated by depolarization but more slowly than Na^+ channel; inactivated as function of cytoplasmic $[Ca^{2+}]$ or V_m	Verapamil, D600, Co^{2+} , Cd^{2+} , Mn^{2+} , Ni^{2+} , La^{3+}	Produces slow depolarization; allows Ca^{2+} to enter cell, where it can act as second messenger
Voltage-gated K^+ channel ("delayed rectifier")	$I_{K(V)}$	Activated by depolarization but more slowly than Na^+ channel; inactivated slowly and not completely if V_m remains depolarized	Intra- and extracellular TEA, amino pyridines	Carries current that rapidly repolarizes the membrane to terminate an AP
Ca^{2+} -dependent K^+ channel	$I_{K(Ca)}$	Activated by depolarization plus elevated cytoplasmic $[Ca^{2+}]$; remains open as long as cytoplasmic $[Ca^{2+}]$ is higher than normal	Extracellular TEA	Carries current that repolarizes the cell following APs based on either Na^+ or Ca^{2+} and that balances I_{Ca} , thus limiting depolarization by I_{Ca}

Osmotic Properties of Cells and Relative Ion Concentrations



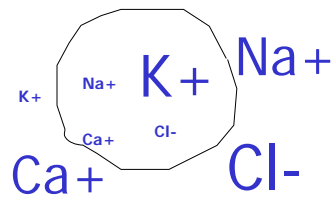
How do we calculate the value of an individual **equilibrium potential**, or the **resting potential** of a cell?

Equilibrium potential (E_x in mV)

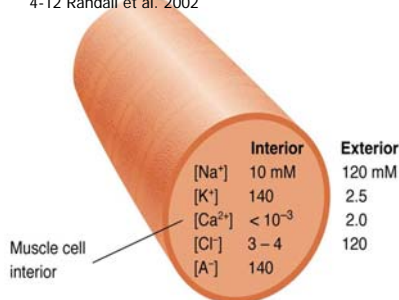
“Every ion’s goal in life is to make the membrane potential equal its own equilibrium potential (E_x in mV)”

53

Osmotic Properties of Cells and Relative Ion Concentrations



4-12 Randall et al. 2002



Permeabilities

$K^+ \gg Na^+ ; Cl^-$

[A⁻] = molar equivalent of negative charges carried by other molecules and ions.

A⁻ (includes proteins, phosphate groups, etc.)

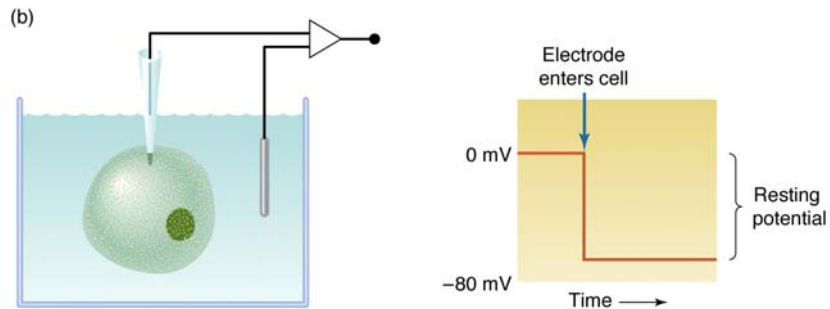
54

At Rest

Membrane Potential
(V_m in volts or mV)

- outside is zero by convention

- V_{rest} K^+, Na^+ about -60 mV



5-7 Randall et al. 2002

55

Nernst equation:
$$E = \frac{RT}{zF} \ln \frac{C_{out}}{C_{in}}$$

where

E = equilibrium membrane potential

R = gas constant

T = absolute temperature

z = valence

F = Faraday's constant

(Mistake in Hill et al. text bottom of page 291; see if you can fix it)

Only in my book?

56

Equilibrium Potential

- Calculate for a given type of ion using the simplified Nernst Equation:

$$E_x = \frac{0.058}{z} \log \frac{[X]_{out}}{[X]_{in}}$$

$$E_{Na} = \frac{0.058}{z} \log \frac{[Na^+]_{out}}{[Na^+]_{in}}$$

$$E_{Na} = \frac{0.058}{1} \log \frac{120 \text{ mM}}{10 \text{ mM}} = 63 \text{ mV} (0.063 \text{ V})$$

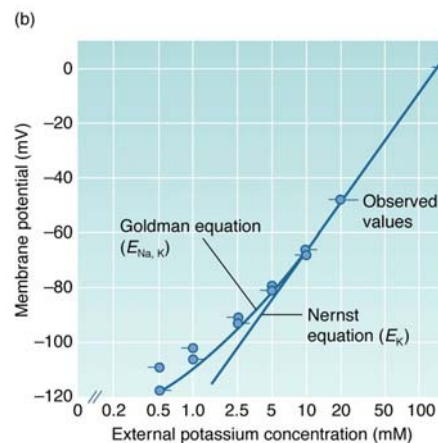
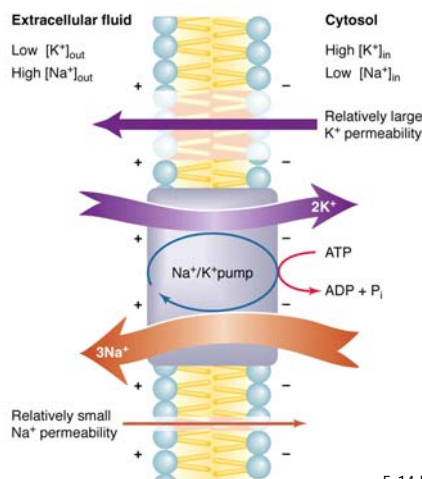
remember Equilibrium potential (E_x in mV)
when [X] gradient = electrical gradient

57

Membrane Potential

To calculate:

- Nernst for single ion
 $V_m = E_x$ if only one ion 'driving'
- Goldman equation for multiple ions



5-14 Randall et al. 2002

58

Nernst Question

Calculate E_K if -101 mV
 $[K^+]_{\text{inside}} = 140 \text{ mM}$
 $[K^+]_{\text{outside}} = 2.5 \text{ mM}$

If the resting membrane potential is -60 mV , which way will K^+ 'want' to move (in or out of the cell)?

OUT

Which way will Na^+ want to move?

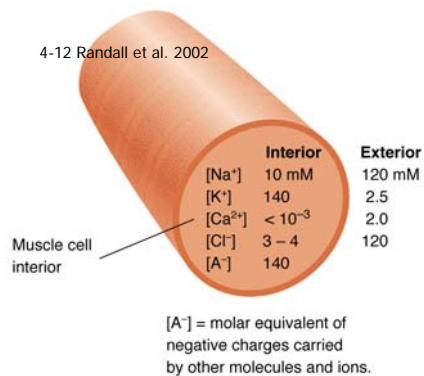
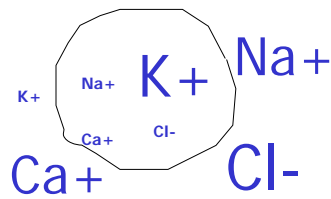
IN

Which way will K^+ want to move if membrane potential is -110 mV ? 30 mV ?

IN OUT

59

Osmotic Properties of Cells and Relative Ion Concentrations



Goldman Equation?

Donnan Equilibrium?

60