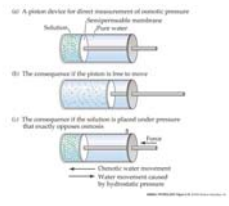
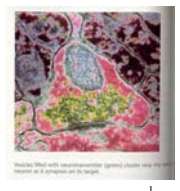


Lecture 5, 28 Jan 2008
 Vertebrate Physiology
 ECOL 437 (MCB/VetSci 437)
 Univ. of Arizona, spring 2008
 Kevin Bonine & Kevin Oh



1. Water, Solutes, Osmosis (Ch 3)
2. Intro Nervous System Fxn (Ch10)



http://eebweb.arizona.edu/eeb_course_websites.htm

Housekeeping, 28 January 2008

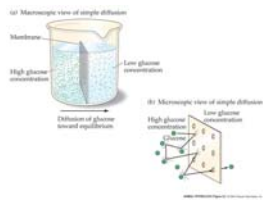


Upcoming Readings

today: Ch 3&10
 Wed 30 Jan: Ch 10&11
 LAB Wed 30 Jan: Bisbal & Specker, plus two optional papers
 (see website for links to papers; "worksheet" via email)
 Fri 01 Feb: Ch11

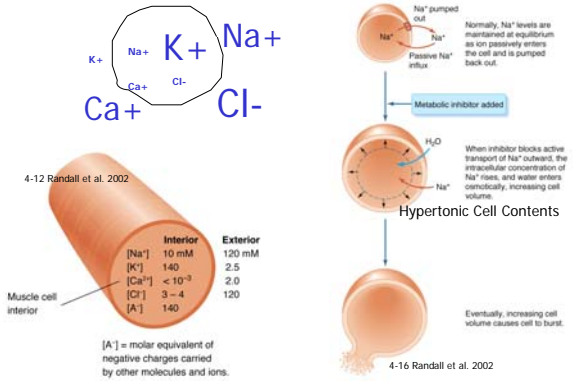
Lab discussion leaders: 30 Jan 1pm – Josh, Seth
 3pm – Aaron, Adam
 Lab discussion leaders: 06 Feb 1pm – Rittner, Whitney
 3pm – Roxanne, Maria

Vertebrate Physiology 437



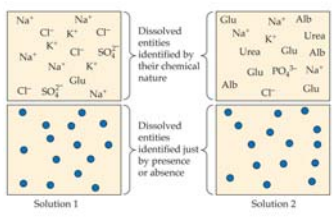
Chapter 3
 Movement of Solutes and Water

Osmotic Properties of Cells and Relative Ion Concentrations

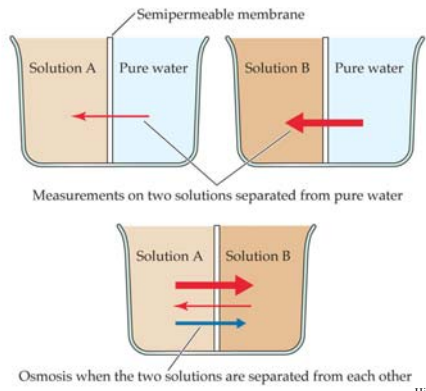


Colligative Properties

- Osmotic Pressure
- Freezing Point
- Water Vapor Pressure (boiling point; evaporation)



Hill et al 2004



Osmosis when the two solutions are separated from each other

Hill et al 2004

$$6 \times 10^{23}$$

Osmolarity

1 osmolar solution (Osm)

has 1 Avogadro's number of dissolved particles/liter solvent

1 milliosmolar solution (mOsm)

has 0.001 Avogadro's number of dissolved particles/liter solvent

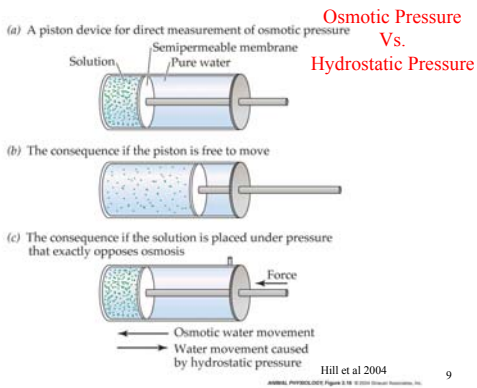
What osmolarity do you get if you add 6×10^{23} molecules of **glucose** to a liter of water?

What osmolarity do you get if you add 6×10^{23} molecules of table **salt** to a liter of water?

NaCl (strong electrolyte)

7

8



9

$$\text{Rate of Osmosis} = K \frac{\Pi_1 - \Pi_2}{X}$$

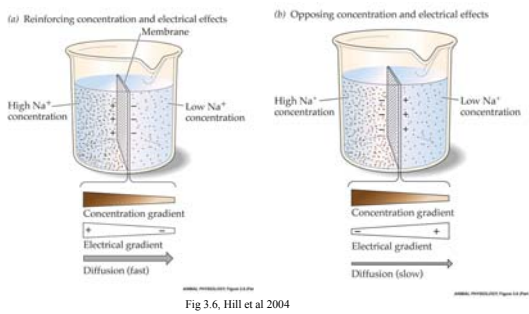
Difference in osmotic potential

Proportionality Coefficient (~ permeability and temp)

Distance between solutions

10

Electrochemical equilibrium



11

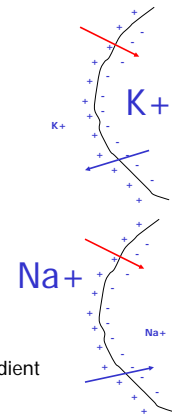
Movement Across Membranes

Electrochemical Gradient

Electrical gradient
Concentration gradient

Electrochemical equilibrium

Equilibrium potential (E_x in mV)
when $[X]$ gradient = electrical gradient

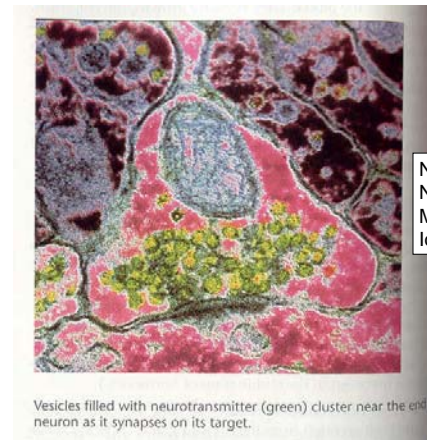


12

Equilibrium potential (E_x in mV)

“Every ion’s goal in life is to make the membrane potential equal its own equilibrium potential (E_x in mV)”

13



p. 214, Silverthorn
2001, 2nd ed.
Human Physiology
Prentice Hall

Nervous System
Neurons
Membranes
Ions

Vesicles filled with neurotransmitter (green) cluster near the end of a neuron as it synapses on its target.

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Nervous System

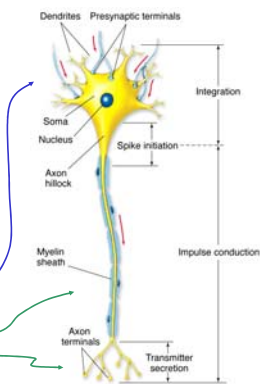
Comprises

- Neurons / Nerve Cells
- Glial Cells (support)

- Signalling via combination of **Electrical and Chemical**

- Integrate information
AFFERENT

- Coordinate Response
EFFERENT



5-2 Randall et al. 2002

Organization of the Nervous System

Three main functions:

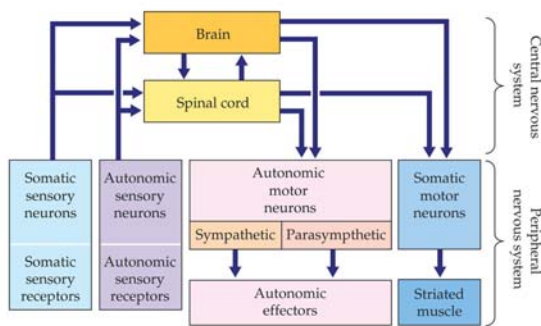
1. **Sensory Reception** (converts environmental stimulus to elect/chem)
2. **Central Processing**
3. **Motor Output**

Divided into CNS and PNS

A. CNS = **Central Nervous System**
- Brain and Spinal Cord
(and eyes and interneurons)

B. PNS = **Peripheral Nervous System**
- most sensory and motor axons

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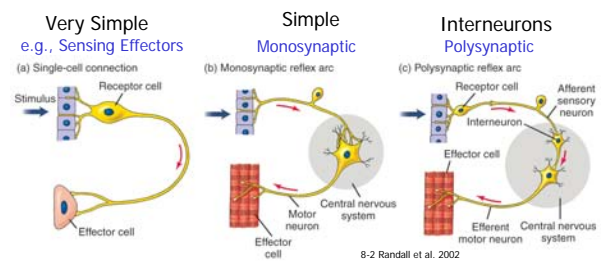


Hill et al. 2004, Fig 10.7

ANIMAL PHYSIOLOGY, Figure 10.7 © 2004 Sinauer Associates, Inc.

Flow of Information

Afferent Signal -> CNS -> Efferent Signal -> Response

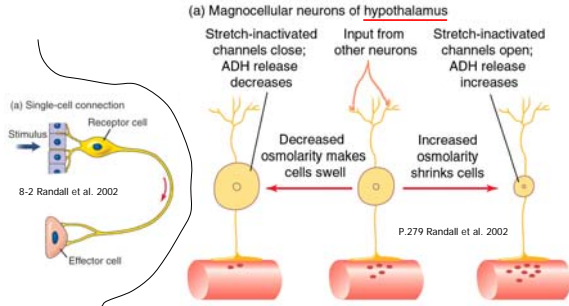


8-2 Randall et al. 2002

Sensing Effectors

Sensors with **Afferent and Efferent Properties/Homeostasis**

e.g., osmolarity and antidiuretic hormone (ADH)



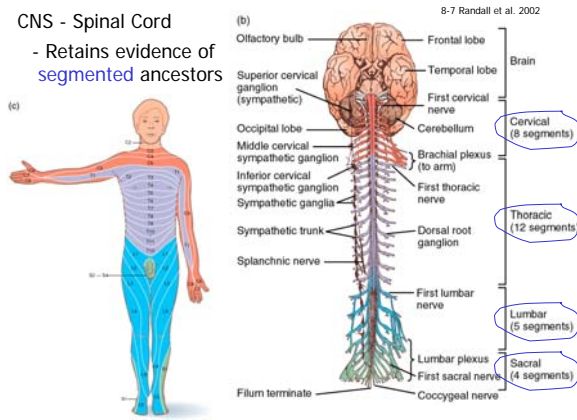
Evolution of Nervous System:

- Based on the neuron
- Elaboration of Reflex Arc
- Group neurons into CNS
- More neurons in complex organisms
- New structures added on to old (not replaced)
- Size of CNS region correlated with importance
- Topological Maps

20

CNS - Spinal Cord

- Retains evidence of segmented ancestors



CNS - Spinal Cord

Functional Anatomy

White matter = myelin
Gray matter = somata and dendrites

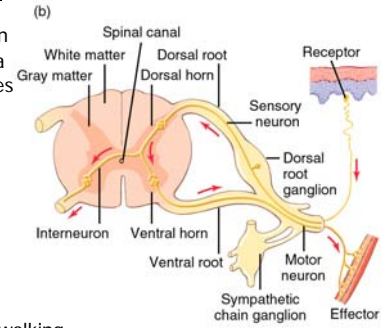
Cerebrospinal Fluid in spinal canal

Dorsal root, horn = -afferent
Ventral root, horn = -efferent

Spinal Reflexes:

- locomotion / walking
- chicken w/o head

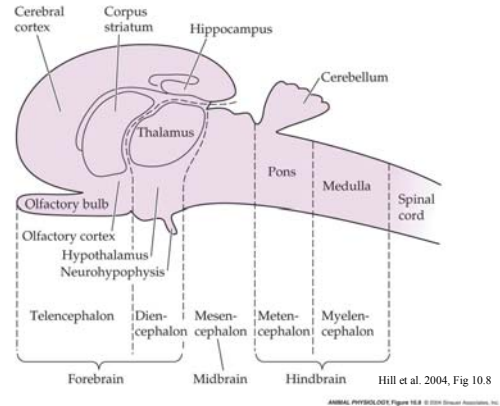
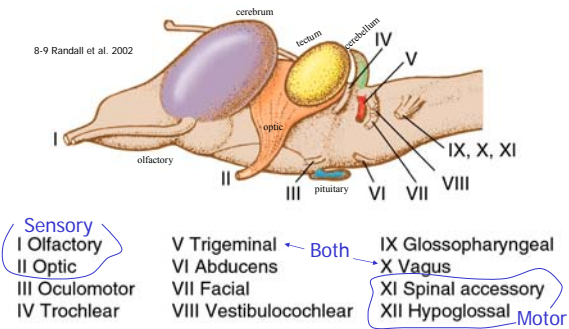
Dorsal Root Ganglion (PNS)
- afferent sensory somata



8-8 Randall et al. 2002 22

CNS - Brain

Vertebrate **bilaterally paired nerve** connections to periphery



CNS - Brain **Functional Anatomy**

-Medulla oblongata

Respiration, autonomic funct, some sensory (hearing, equil.)

-Cerebellum

Coordinate motor output

Integrates info. from proprioceptors (stretch and joint) visual, auditory

More convoluted (↑ s.a.) in higher groups

Birds with large cerebellum to handle 3D flight

-Pons (and tectum)

Integrate and communicate

Visual, tactile, auditory maps

~ body movement coordination in some groups

-Cerebral Cortex

In higher groups takes over function of tectum

25

CNS - Brain **Functional Anatomy (con't)**

-Thalamus

Sensory and motor coordination

Often communicates with cerebral cortex

-Amygdala

Processes info. and output related to emotions

-Hypothalamus

Also involved in emotions

Body temp, eating, drinking, sex

Water and electrolyte balance

-Olfactory Bulb

Key sense in many vertebrate groups

Anterior position

-Cerebrum (covered by cerebral cortex)

More evolved in higher groups (size and folds)...

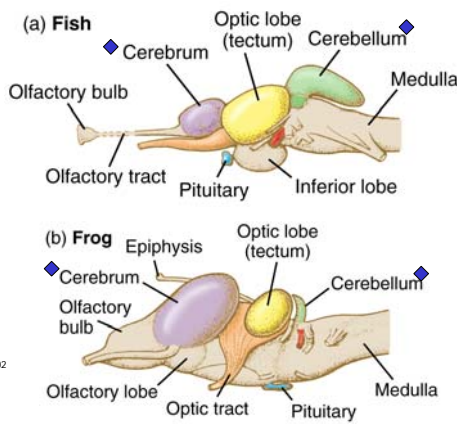
Olfaction straight to cerebrum w/o going through thalamus (in mammals)

26

CNS - Brain

◆ Note change in size in diff. groups

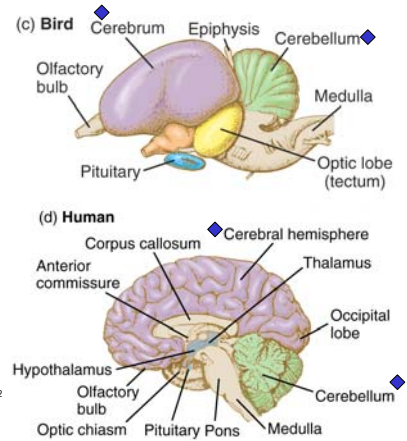
8-10 Randall et al. 2002



CNS - Brain

◆ Note change in size in diff. groups

8-10 Randall et al. 2002



CNS - Brain

Cerebrum and Cerebral Cortex

- Folds increase surface area and # neurons

- Functional Regions

1. Sensory cortex

- somatosensory, auditory, visual
- sensory homunculus ("little man")

2. Motor cortex

- often similar to sensory cortex map

3. Association cortex

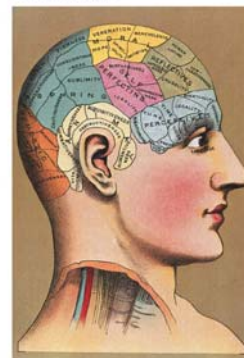
- memory, future, thought, communication

Relative importance of each region changes among vertebrate groups

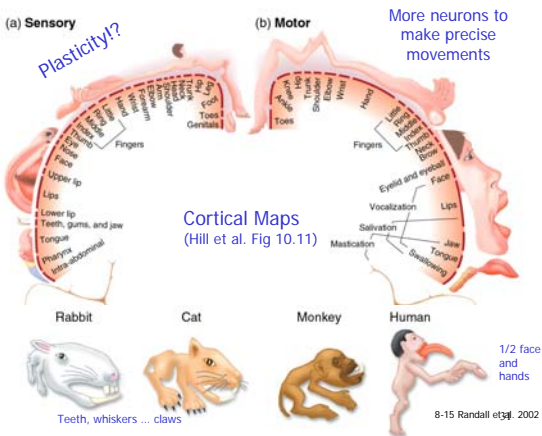
Size of receptive fields?

29

(a) Phrenology



Hill et al. 2004, Fig 10.9
30



Structural and Functional Regions

CNS

- Most neuronal somata incl. motor neurons
- Interneurons

Nuclei = collections of somata w/ similar function

Tracts = bundles of axons from nuclei

PNS

- Nervous system outside CNS

Nerves = axon bundles from sensory + motor neurons

Ganglia = somata of some autonomic neurons and most sensory neurons

Nerve usually with both Afferent and Efferent axons

32

Structural and Functional Regions

8-6 Randall et al. 2002

- Efferent NS

1. Somatic/Voluntary -skeletal muscle

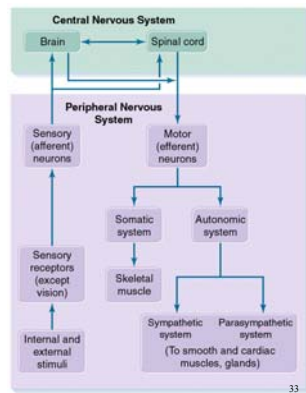
2. Autonomic

- smooth muscle
- cardiac muscle
- glands
- "housekeeping"

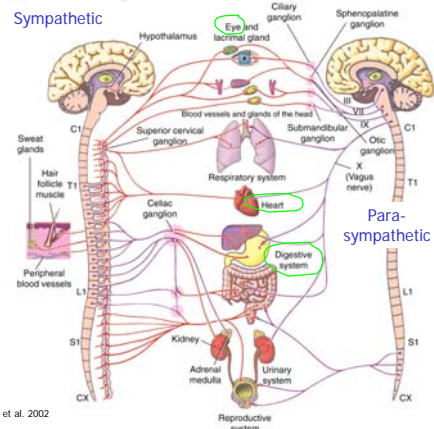
A. Sympathetic ~ fight or flight

B. Parasympathetic ~ rest and digest

C. Enteric



33



8-18 Randall et al. 2002

34

Autonomic NS (vs Voluntary/Somatic)

Antagonistic Groups in Balance:

- A. Sympathetic (f or f)
- B. Parasympathetic (r + d)

Both function via reflex arcs, but often opposite effects

Efferent signal with two neurons:

1. Preganglionic (NT released is Acetylcholine [Ach])
2. Postganglionic (PNS, receptor is nicotinic Ach)

Difference between Symp. and Para. is in:

1. CNS origin
2. Location of postganglionic somata
3. Postganglionic NT
4. Receptors on target tissues

35

Autonomic NS

Sympathetic

- 2-Postganglionic somata nearer CNS in chain ganglia
- 3- Postganglionic NT is Norepinephrine
- 4-Effector receptor is alpha or beta adrenergic

Parasympathetic

- 2-Postganglionic somata near effector, or in effector organ
- 3- Postganglionic NT is Ach
- 4-Effector receptor is muscarinic Ach

Difference between Symp. and Para. is in:

1. CNS origin
2. Location of postganglionic somata
3. Postganglionic NT
4. Receptors on target tissues

36

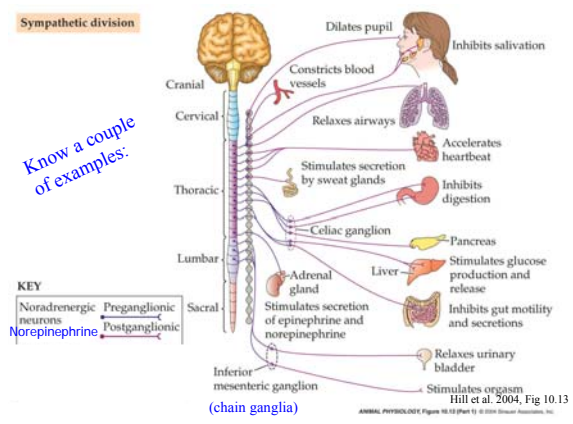
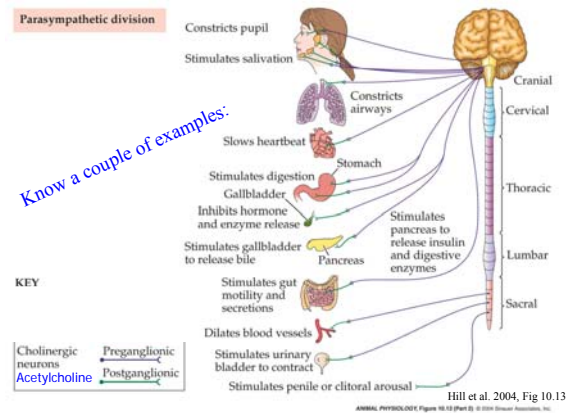
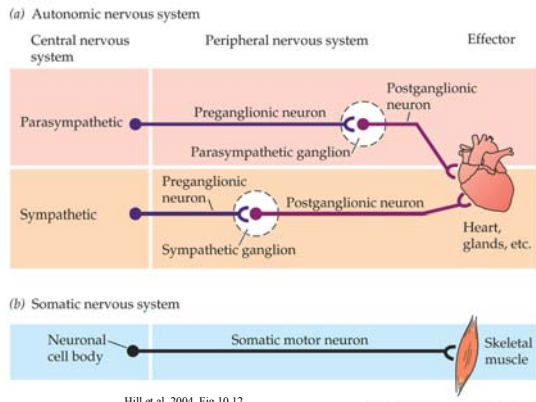


TABLE 10.2 Major actions of sympathetic and parasympathetic divisions in vertebrates

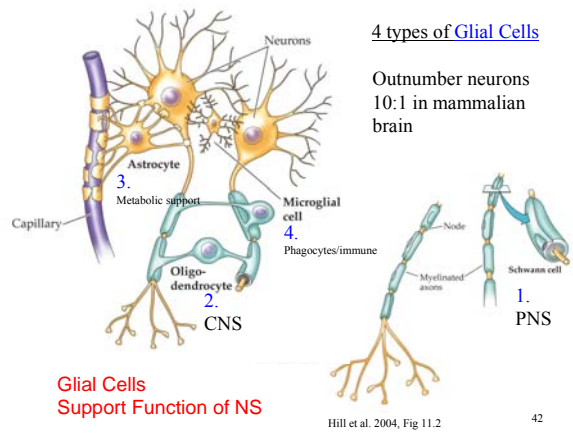
Process	Parasympathetic effect	Sympathetic effect
Digestion: gastrointestinal secretion and motility	Stimulates	Inhibits
Heartbeat	Slows	Increases rate and force
Blood vessels	Usually dilates	Constricts vessels to kidneys and gut; dilates vessels to skeletal muscles
Blood pressure	Decreases	Increases
Lung passages	Constricts	Dilates
Secretion of epinephrine and norepinephrine by adrenal medullary glands	—	Stimulates

Hill et al. 2004

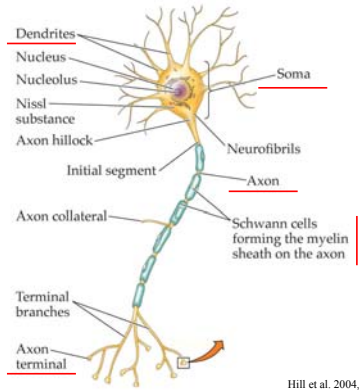


“Squid axons are important to physiologists, and to the squid.”
Hill et al. 2004, p.281

Sir Alan Hodgkin, Nobel Prize 1963



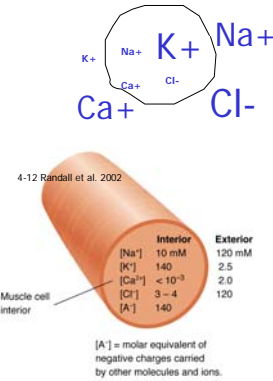
Neurons:



Hill et al. 2004, Fig 11.1

ANIMAL PHYSIOLOGY: Figures 11.1 (Part 1) © 2004 Sinauer Associates, Inc.

Osmotic Properties of Cells and Relative Ion Concentrations



To understand how the NS works we need to return to [Membrane Details](#)

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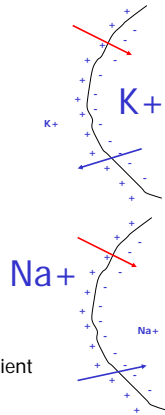
Movement Across Membranes

Electrochemical Gradient

Electrical gradient
Concentration gradient

Electrochemical equilibrium

Equilibrium potential (E_x in mV)
when [X] gradient = electrical gradient



45

Equilibrium potential (E_x in mV)

“Every ion’s goal in life is to make the membrane potential equal its own equilibrium potential (E_x in mV)”

46

Membrane Potential

1. To change V_m :
A Small Number of Ions actually move relative to the number present both inside and outside the cell
2. Concentration gradients...
(previously established by ATPase pumps) are **not abolished** when the channels for an ion species open

[Gradients allow for ‘work’ to be done, e.g., action potential sends signal along axon]

47

Membrane Potential

3. Driven by ions that are permeable to the membrane (and have different []_{in} as compared to []_{out} a.k.a. gradient created with ATP) - K⁺ for example
4. Equilibrium Potential (E_x in mV):
~ The equilibrium potentials of all the permeable ions (a function of their established gradients) will determine the membrane potential of a cell
5. emf determines which direction a given ion (X) will move when the membrane potential is known

$$emf_x = V_m - E_x$$

48

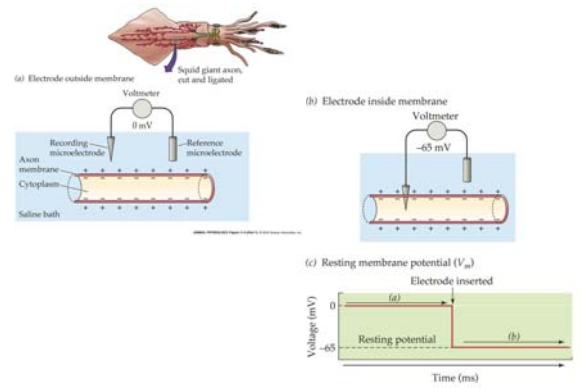
Membrane Potential

6. Resting Membrane Potential driven by K^+ efflux and, to a lesser extent, Na^+ influx

7. Na^+/K^+ ATPase pump generates gradients that, for these permeable ions, determine membrane potential



49



Hill et al. 2004, Fig 11.4

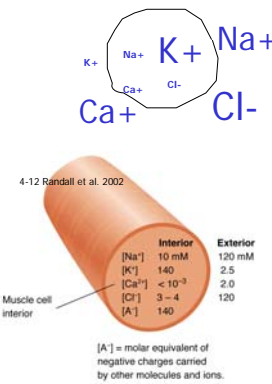
50

Table 5-1 Examples of ion channels found in axons

Randall et al. 2002

Channel	Current through channel	Characteristics	Selected blockers	Function
Leak channel (open in resting axon)	I_K (leak)	Produces relatively high I_K of resting cell	Partially blocked by tetraethylammonium (TEA)	Largely responsible for V_{rest}
Voltage-gated Na^+ channel	I_{Na}	Rapidly activated by depolarization; becomes inactivated even if V_m remains depolarized	Tetrodotoxin (TTX)	Produces rising phase of AP
Voltage-gated Ca^{2+} channel	I_{Ca}	Activated by depolarization but more slowly than Na^+ channel; inactivated as function of cytoplasmic $[Ca^{2+}]$ or V_m	Verapamil, D900, Co^{2+} , Gd^{3+} , Mn^{2+} , Ni^{2+} , La^{3+}	Produces slow depolarization; allows Ca^{2+} to enter cell, where it can act as second messenger
Voltage-gated K^+ channel ("delayed rectifier")	$I_{K(D)}$	Activated by depolarization but more slowly than Na^+ channel; inactivated slowly and not completely if V_m remains depolarized	Intra- and extracellular TEA, aminopyridines	Carries current that rapidly repolarizes the membrane to terminate an AP
Ca^{2+} -dependent K^+ channel	$I_{K(Ca)}$	Activated by depolarization plus elevated cytoplasmic $[Ca^{2+}]$; remain open as long as cytoplasmic $[Ca^{2+}]$ is higher than normal	Extracellular TEA	Carries current that repolarizes the cell following APs based on either Na^+ or Ca^{2+} and that balances I_{Ca} , thus limiting depolarization by I_{Ca}

Osmotic Properties of Cells and Relative Ion Concentrations



How do we calculate the value of an individual equilibrium potential, or the resting potential of a cell?

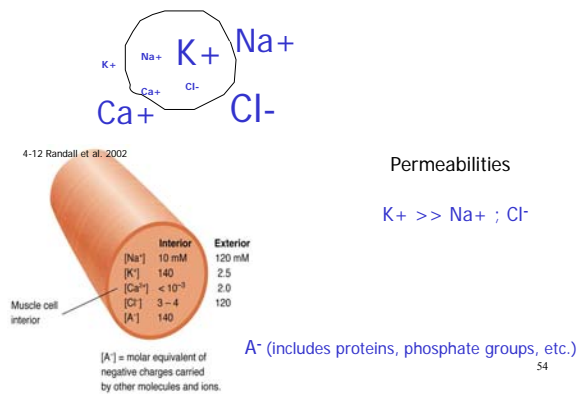
52

Equilibrium potential (E_x in mV)

"Every ion's goal in life is to make the membrane potential equal its own equilibrium potential (E_x in mV)"

53

Osmotic Properties of Cells and Relative Ion Concentrations

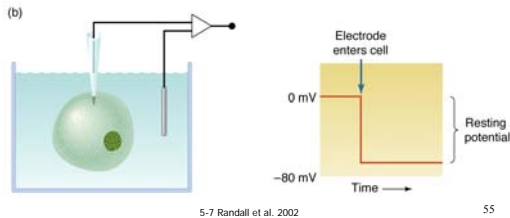


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At Rest

Membrane Potential
(V_m in volts or mV)

- outside is zero by convention
- V_{rest} K^+, Na^+ about -60 mV



Nernst equation:
$$E = \frac{RT}{zF} \ln \frac{C_{out}}{C_{in}}$$

where
 E = equilibrium membrane potential
 R = gas constant
 T = absolute temperature
 z = valence
 F = Faraday's constant

(Mistake in Hill et al. text bottom of page 291; see if you can fix it)
 Only in my book?

Equilibrium Potential

- Calculate for a given type of ion using the simplified Nernst Equation:

$$E_x = \frac{0.058}{z} \log \frac{[X]_{out}}{[X]_{in}}$$

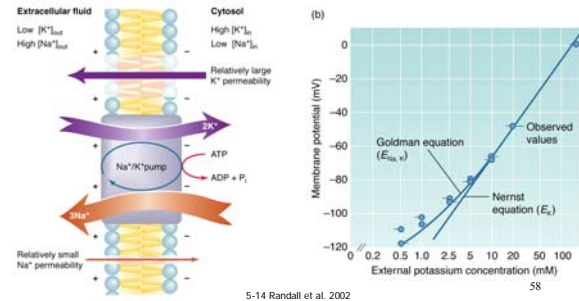
$$E_{Na} = \frac{0.058}{z} \log \frac{[Na^+]_{out}}{[Na^+]_{in}}$$

$$E_{Na} = \frac{0.058}{1} \log \frac{120 \text{ mM}}{10 \text{ mM}} = 63 \text{ mV} (0.063 \text{ V})$$

remember Equilibrium potential (E_x in mV)
 when [X] gradient = electrical gradient

Membrane Potential

- To calculate:
 - Nernst for single ion $V_m = E_x$ if only one ion 'driving'
 - Goldman equation for multiple ions



Nernst Question

Calculate E_K if -101 mV
 $[K^+]_{inside} = 140 \text{ mM}$
 $[K^+]_{outside} = 2.5 \text{ mM}$

If the resting membrane potential is -60 mV , which way will K^+ 'want' to move (in or out of the cell)?

OUT

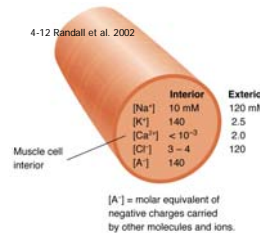
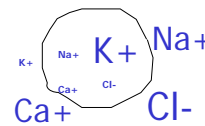
Which way will Na^+ want to move?

IN

Which way will K^+ want to move if membrane potential is -110 mV ? 30 mV ?

IN OUT

Osmotic Properties of Cells and Relative Ion Concentrations



Goldman Equation?

Donnan Equilibrium?