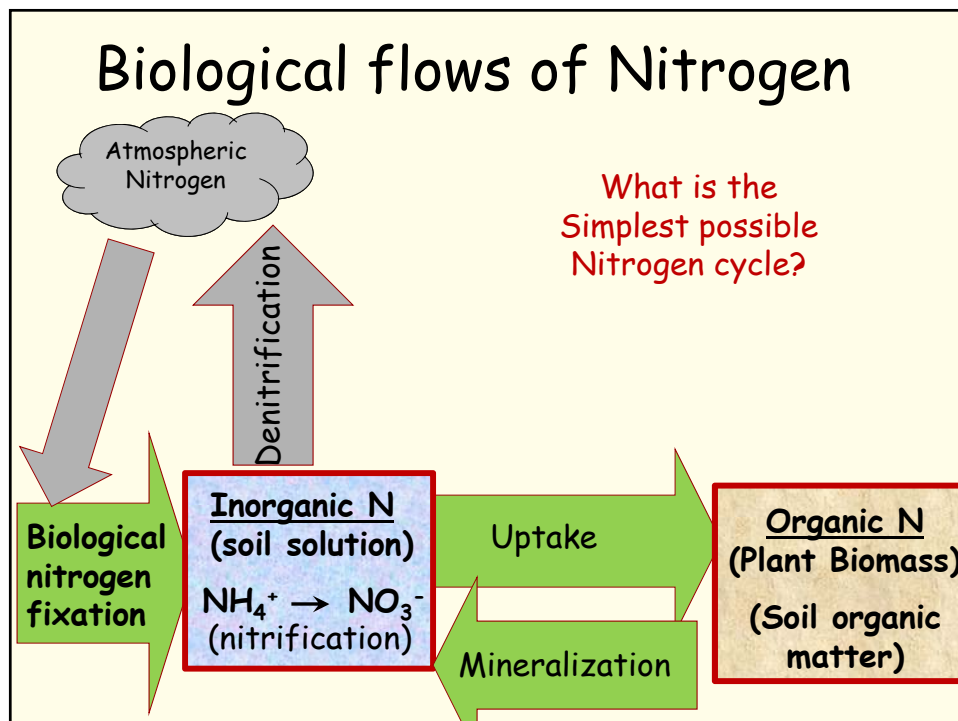


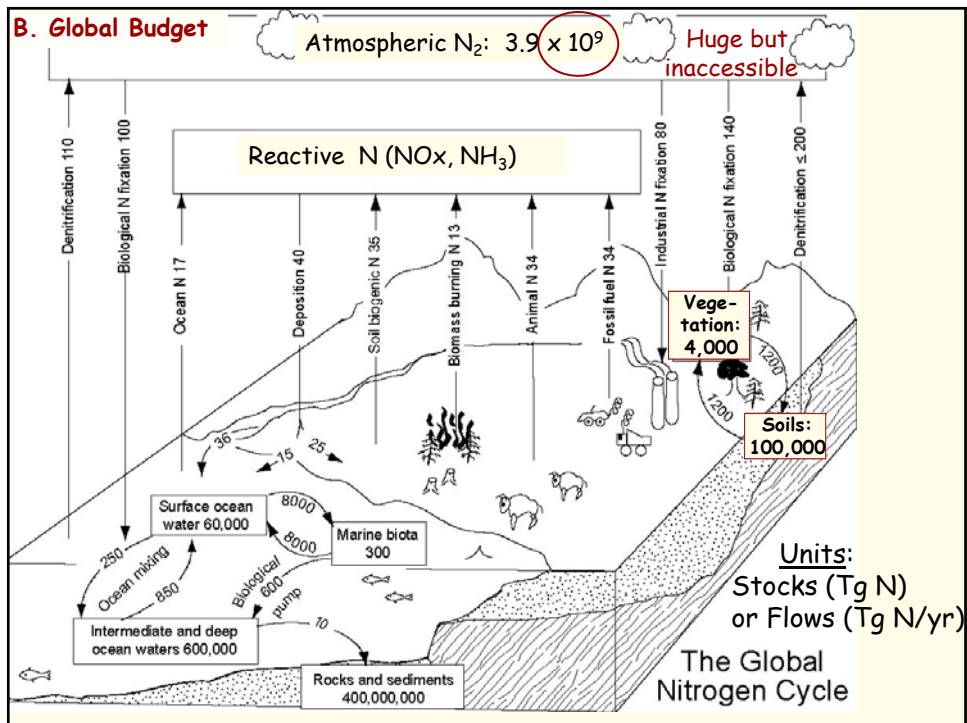
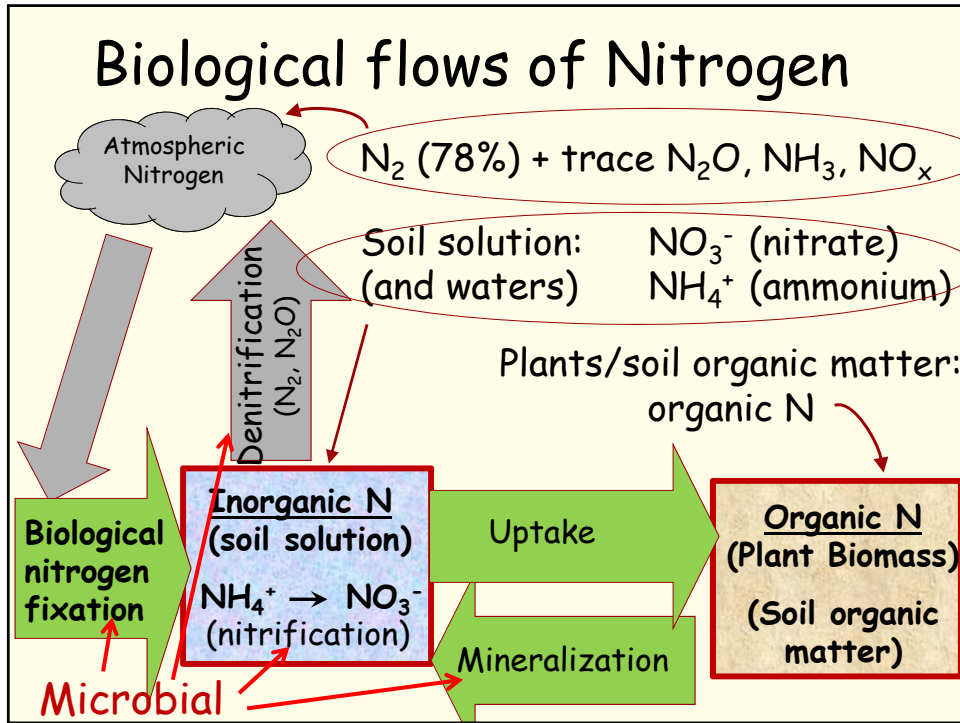
N-cycle: biogeochemistry

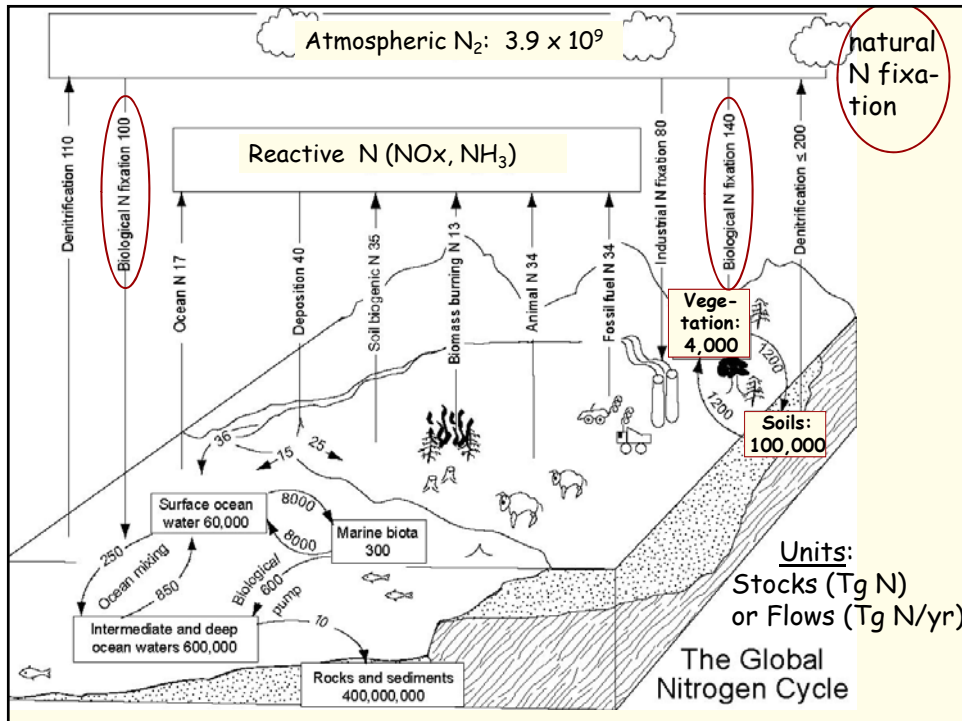
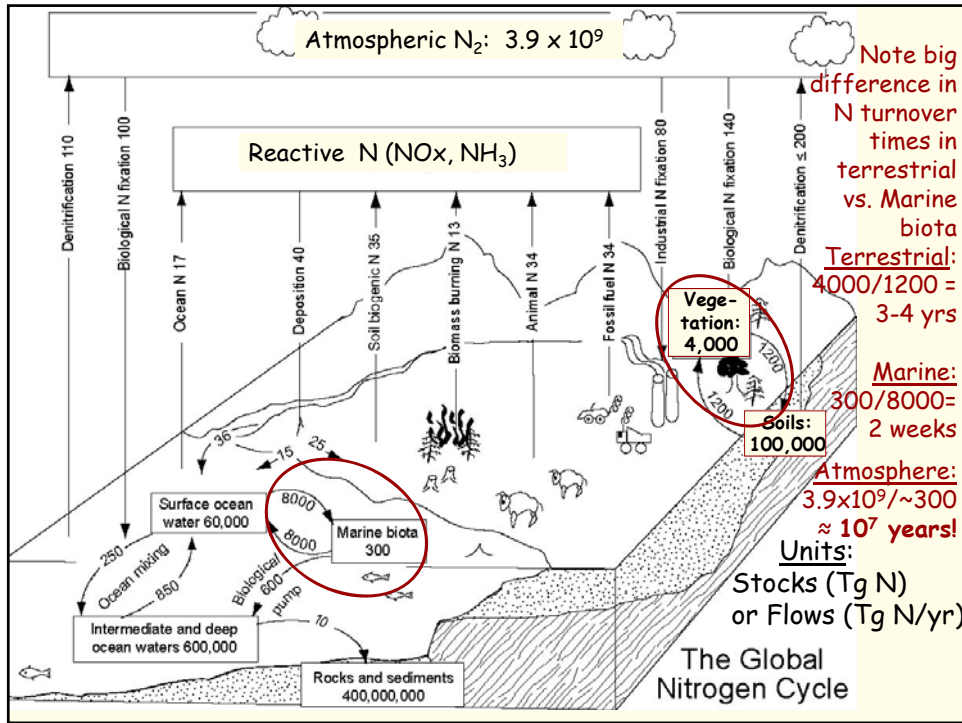
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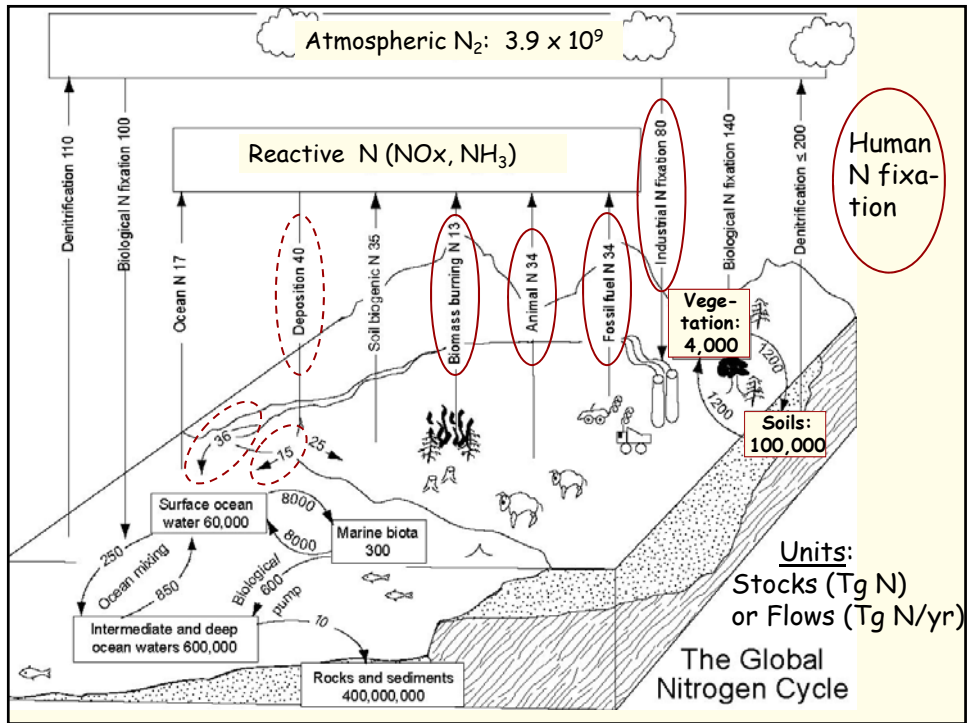
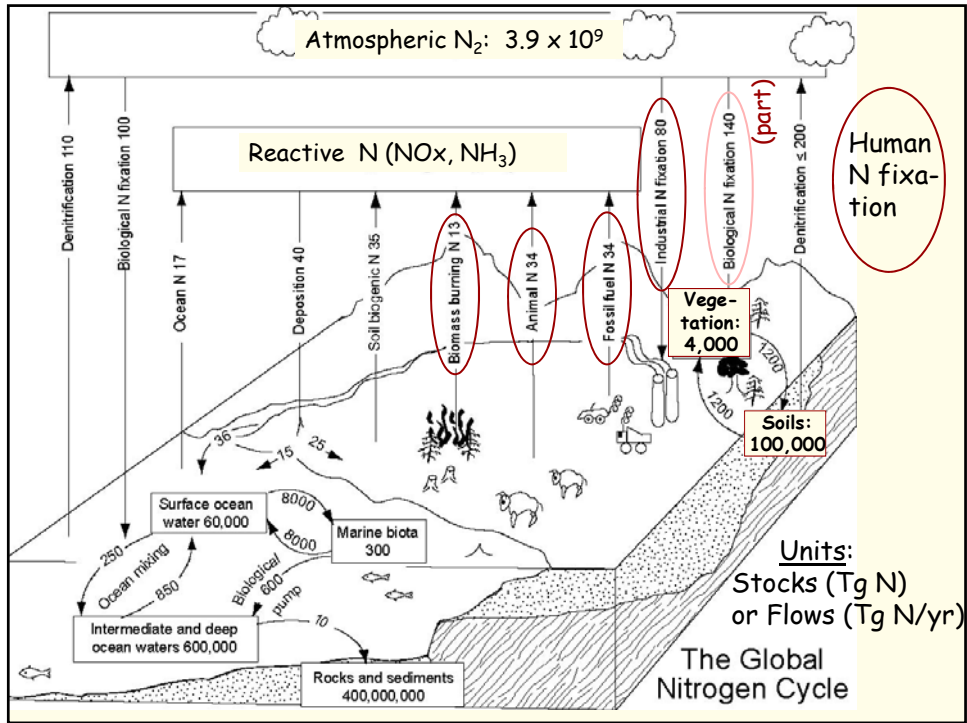
April 4, 2014

- I. N cycling
 - A. simplest possible
 - B. Global N budget
 - C. Effects of N-cycling ('the Nitrogen Cascade')
- II. Nitrous Oxide (N₂O) budgets
- III. Data-Model Intercomparison studies of C-cycle feedbacks to climate





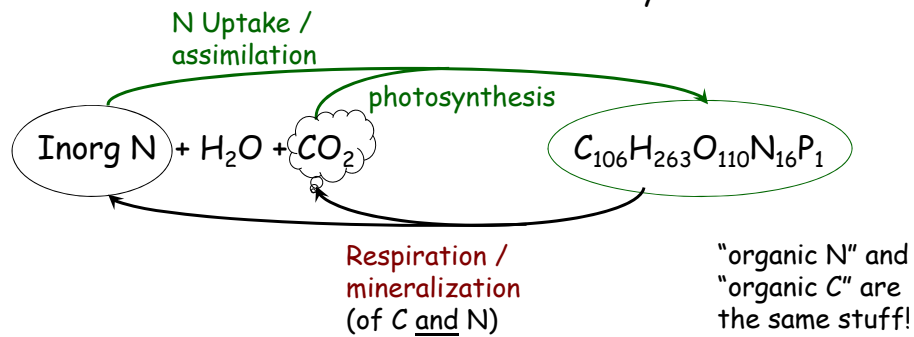




Stoichiometry (proportional elemental composition, e.g. C:N:P in biomass)

(because biogeochemical cycles do not exist in isolation, but are coupled)

Example: Integrated N-C cycle



Stoichiometry (proportional elemental composition, e.g. C:N:P in biomass)

Stoichiometry:

Greek: *stoikheion* (element)
+ *metron* (measure)

ecological stoichiometry: definite proportions of elemental combinations, a **constraint** on biogeochemistry of life, from the cell to the organism to the biosphere

(broader concept than the stoichiometry of balancing a chemical reaction)

Stoichiometry (proportional elemental composition, e.g. C:N:P in biomass)

Originally introduced as an ecological concept by A. Redfield in 1934 in his study of marine phytoplankton. the *Redfield ratio*, $C:N:P = 106:16:1$
(relative amounts of C, N, & P in phytoplankton)

Based on what we know now about marine v terrestrial residence times (and other things we know about these systems), what might we infer/guess about terrestrial vs. marine C:N:P ?

Stoichiometry (proportional elemental composition, e.g. C:N:P in biomass)

Originally introduced as an ecological concept by A. Redfield in 1934 in his study of marine phytoplankton. the *Redfield ratio*, $C:N:P = 106:16:1$
(relative amounts of C, N, & P in phytoplankton)

Much more recently applied to terrestrial systems:
McGroddy et al. 2004
 Forest Foliage C:N:P = 1200:28:1 (Global scale)
 Litter = 3000:45:1 (resorption)
 (higher C than in marine phytoplankton: reflecting importance of carbon-rich structural components - e.g. cellulose)

C. Global Change and the N-cycle

The New York Times Search All NYTimes.com

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Beyond Carbon: Scientists Worry About Nitrogen's Effects



“The nitrogen dilemma is not just thinking that carbon is all that matters. But also thinking that global warming is the only environmental issue. The weakening of biodiversity, the pollution of rivers... Smog. Acid rain. Coasts. Forests. It’s all nitrogen.”
- Peter Vitousek

A woman clears away algae in Quindao, China, a buildup caused by too much nitrogen in the Yellow Sea.
By RICHARD MORGAN
Published: September 1, 2008

Correction Appended

TOOLIK FIELD STATION, Alaska — As Anne Giblin was lugging

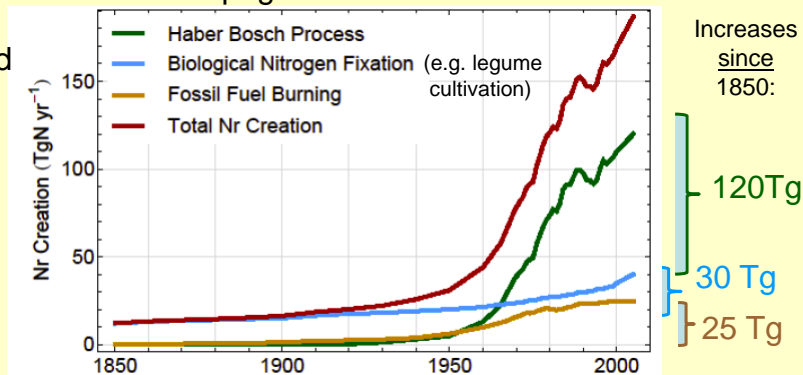
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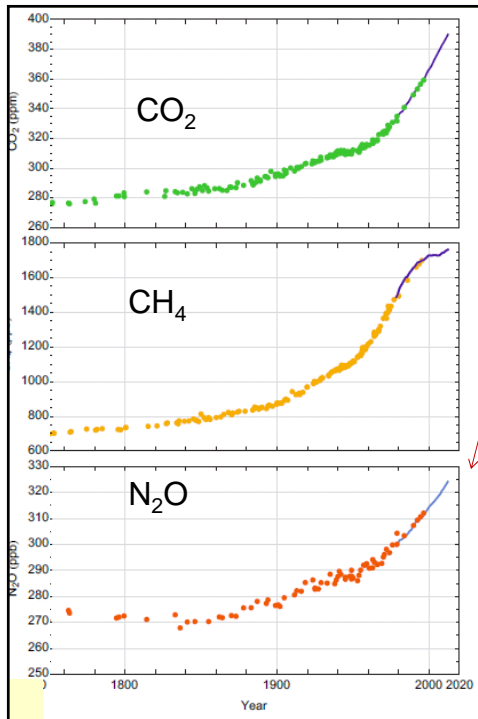
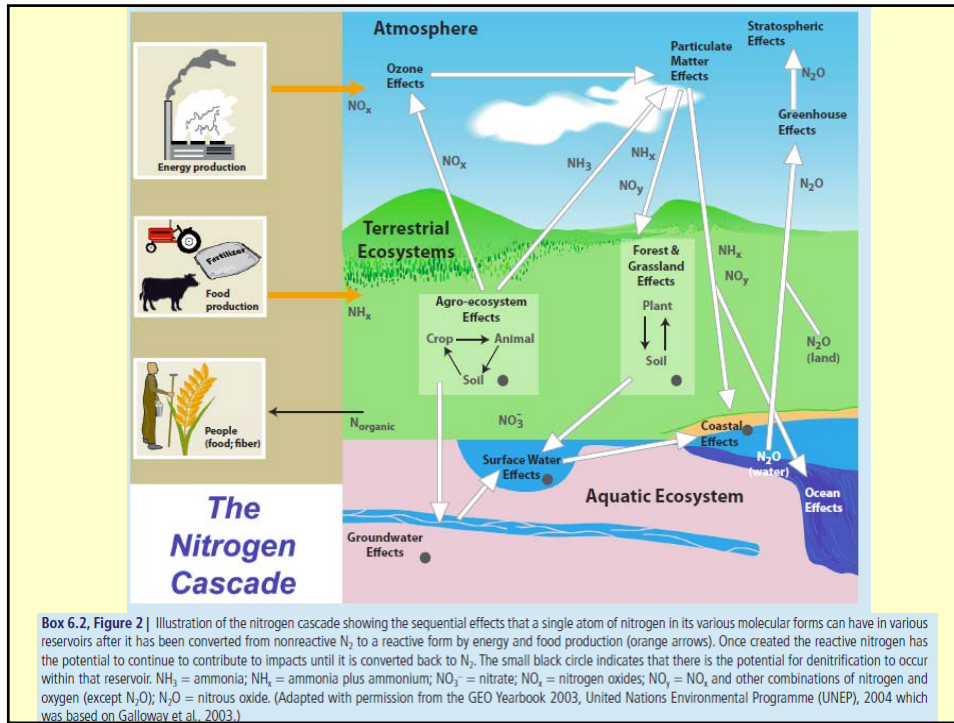
C. Global Change and the N-cycle

Human alteration of terrestrial N-cycle is **LARGE**
(> 100% in fixation relative to the terrestrial background rate of 50-100 Tg N/yr):

No place on earth unaffected by this

Anthropogenic N-fixation





II. Nitrous Oxide Budget

What is causing the increase in atmospheric N_2O ?

N_2O a powerful greenhouse gas, and contributor to O_3 depletion):

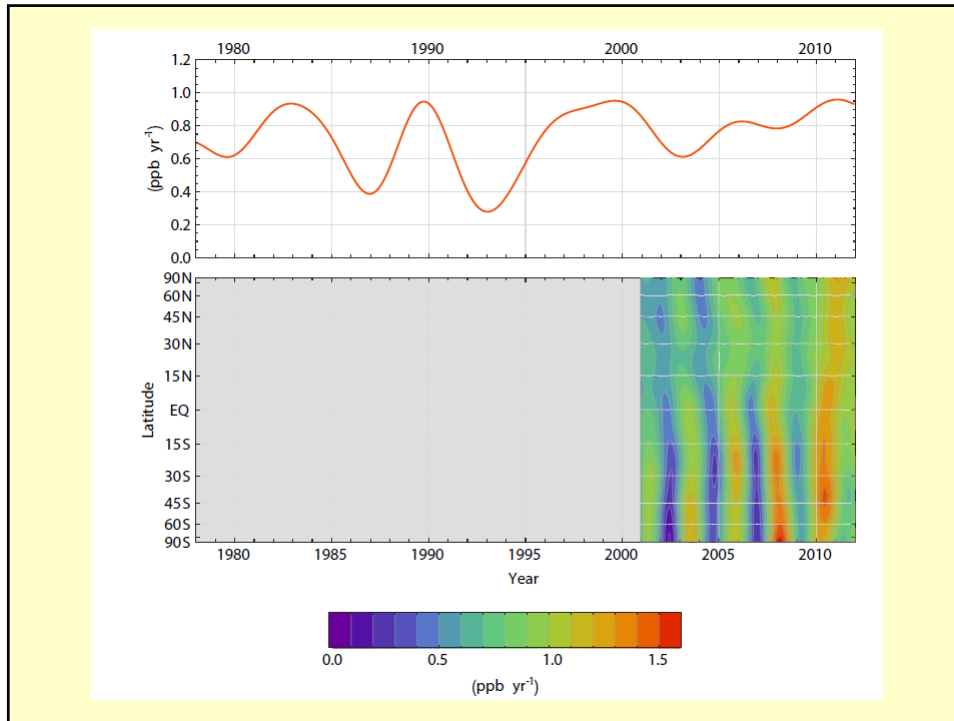
- Increasing 0.25% per year
- Residence time: 120 years
- Greenhouse effectiveness: 300 x CO_2

Sources:

- Natural: microbial denitrification in soils and oceans;
- Anthropogenic: fertilizer production (enhances denitrification), fossil fuel and biomass burning (combustion byproduct).

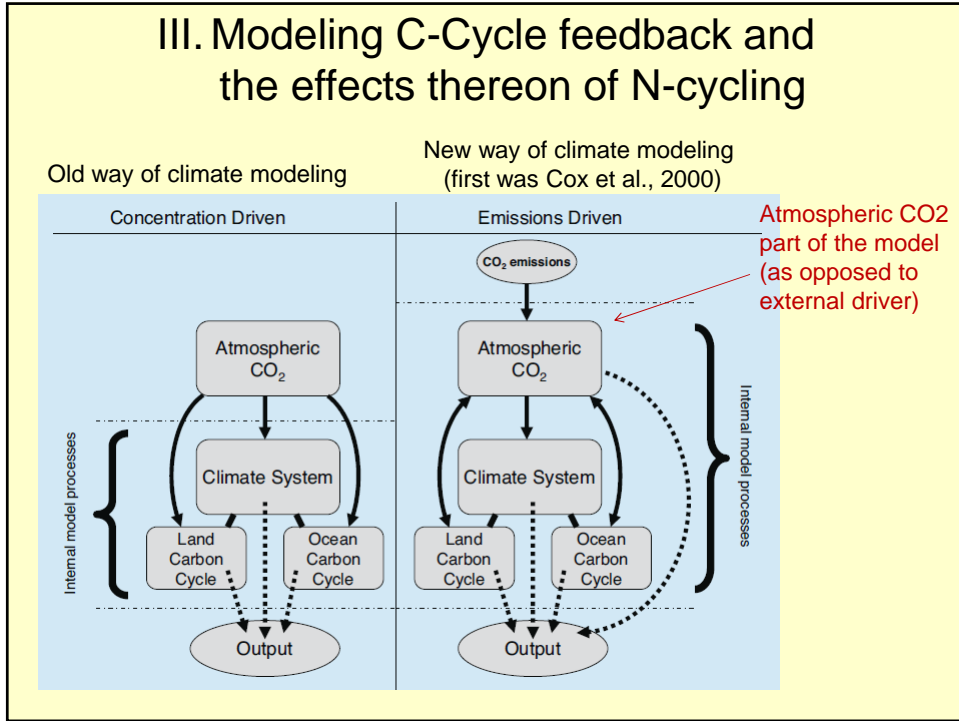
Sink:

- Photolysis by UV light in stratosphere



SECTION 2 (N ₂ O)				
		AR5 (2006/2011)		
2006 N2O budget (IPCC 2013)	Anthropogenic sources		2011 (top-down) Note: microbial!	
	Fossil fuel combustion and industrial processes	0.7 (0.2–1.8) ^p		
	Agriculture	4.1 (1.7–4.8) ^p		
	Biomass and biofuel burning	0.7(0.2–1.0) ^a		
	Human excreta	0.2 (0.1–0.3) ^p		
	Rivers, estuaries, coastal zones	0.6 (0.1–2.9) ^f		
	Atmospheric deposition on land	0.4 (0.3–0.9) ^d		
	Atmospheric deposition on ocean	0.2 (0.1–0.4) ^p		
	Surface sink	-0.01 (0– -1) ^f		
	Total anthropogenic sources	6.9 (2.7–11.1)		6.7 (+/- 1.3)
	Natural sources^g			
	Soils under natural vegetation	6.6 (3.3–9.0)		
	Oceans	3.8(1.8–9.4)		
Lightning	—			
Atmospheric chemistry	0.6 (0.3–1.2)			
Total natural sources	11.0 (5.4–19.6)	9.1 (+/- 1.0)		
Total natural + anthropogenic sources	17.9 (8.1–30.7)	15.8 (+/- 1.0)		
Stratospheric sink	14.3 (4.3–27.2)^g	11.9 (+/- 0.9)		
Observed growth rate	3.61 (3.5–3.8)^h	4.0 (+/- 0.5)		

III. Modeling C-Cycle feedback and the effects thereon of N-cycling



III. Modeling C-Cycle feedback and the effects thereon of N-cycling

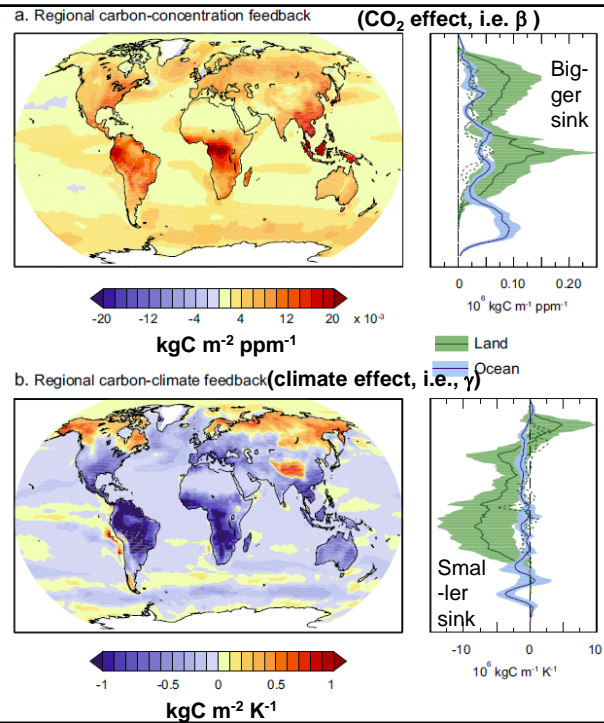
Configuration of simulations for partitioning carbon-climate feedback effects

	CO ₂ input to radiation scheme	CO ₂ input to carbon-cycle scheme	Reason
Fully coupled			Simulates the fully coupled system
effects of ↑ CO ₂ (not including climate)			Isolates the carbon-cycle response to CO ₂ (β) for land and oceans
effects of climate change (not including direct CO ₂)			Isolates carbon-cycle response to climate change (γ) for land and oceans

Model-predicted changes in the fate of anthropogenic CO₂ with climate change

(a. CO₂ effect,
b. Climate effect)

IPCC (2013), Ch. 6



Nitrogen and Carbon dynamics are coupled

Global warming may **accelerate** respiration/mineralization of organic matter,
 → **releasing CO₂** (a positive feedback to warming), but also
 → **releasing inorganic N**, increasing the potential for more carbon uptake
 (a negative feedback to warming)

Integrated N-C cycle
(Focus on terrestrial cycling component)

